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TERMINAL (ENTER 1, 2, 3, OR ?):2

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NEWS 1

Web Page URLs for STN Seminar Schedule - N. America

"Ask CAS" for self-help around the clock

NEWS 3

May 12

EXTEND option available in structure searching

NEWS 4

May 12

Polymer links for the POLYLINK command completed in REGISTRY

NEWS 5

May 27

New UPM (Update Code Maximum) field for more efficient patent

SDIs in CAplus

NEWS 6

May 27

CAplus super roles and document types searchable in REGISTRY

NEWS 7

Jun 28

Additional enzyme-catalyzed reactions added to CASREACT

NEWS 8

Jun 28

ANTE, AQUALINE, BIOENG, CIVILENG, ENVIROENG, MECHENG,

and WATER from CSA now available on STN(R)

PRINCEPPEN expensed with pay display and select options.

NEWS 9 Jul 12 BEILSTEIN enhanced with new display and select options, resulting in a closer connection to BABS

NEWS EXPRESS MARCH 31 CURRENT WINDOWS VERSION IS V7.00A, CURRENT MACINTOSH VERSION IS V6.0c(ENG) AND V6.0Jc(JP), AND CURRENT DISCOVER FILE IS DATED 26 APRIL 2004

NEWS HOURS STN Operating Hours Plus Help Desk Availability

NEWS INTER General Internet Information

NEWS LOGIN Welcome Banner and News Items

NEWS PHONE Direct Dial and Telecommunication Network Access to STN

NEWS WWW CAS World Wide Web Site (general information)

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=> file caplus COST IN U.S. DOLLARS

SINCE FILE TOTAL ENTRY SESSION 0.21 0.21

FULL ESTIMATED COST

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FILE COVERS 1907 - 28 Jul 2004 VOL 141 ISS 5 FILE LAST UPDATED: 27 Jul 2004 (20040727/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> s (meth)acrylic esters or meth-acyrilic ester or meth acrylic ester or (meth)acrylic acid or meth-acrylic acid meth acrylic acid MISSING OPERATOR METH)ACRYLIC
The search profile that was entered contains terms or

The search profile that was entered contains terms or nested terms that are not separated by a logical operator.

=> s (meth)acrylic esters or meth-acyrilic ester or meth acrylic ester
or(meth)acrylic acid or meth-acrylic acid or meth acrylic acid
MISSING OPERATOR METH)ACRYLIC

The search profile that was entered contains terms or nested terms that are not separated by a logical operator.

=> s meth acrylic ester 33548 METH 3 METHS 33550 METH (METH OR METHS) 233395 ACRYLIC 1256 ACRYLICS 233754 ACRYLIC (ACRYLIC OR ACRYLICS) 547111 ESTER 408573 ESTERS 763869 ESTER (ESTER OR ESTERS) 517 METH ACRYLIC ESTER L1(METH (W) ACRYLIC (W) ESTER) => s meth acrylic acid 33548 METH 3 METHS 33550 METH (METH OR METHS) 233395 ACRYLIC 1256 ACRYLICS 233754 ACRYLIC (ACRYLIC OR ACRYLICS) 3846720 ACID

(ACID OR ACIDS)

1439656 ACIDS 4315045 ACID

```
10/657,157
          6529 METH ACRYLIC ACID
                 (METH (W) ACRYLIC (W) ACID)
=> s meth-acrylic acid
         33548 METH
             3 METHS
         33550 METH
                 (METH OR METHS)
        233395 ACRYLIC
          1256 ACRYLICS
        233754 ACRYLIC
                 (ACRYLIC OR ACRYLICS)
       3846720 ACID
       1439656 ACIDS
       4315045 ACID
                 (ACID OR ACIDS)
          6529 METH-ACRYLIC ACID
L3
                 (METH(W)ACRYLIC(W)ACID)
=> s (meth)acrylic acid
MISSING OPERATOR METH) ACRYLIC
The search profile that was entered contains terms or
nested terms that are not separated by a logical operator.
=> dup rem 11 12
PROCESSING COMPLETED FOR L1
PROCESSING IS APPROXIMATELY 16% COMPLETE FOR L2
PROCESSING IS APPROXIMATELY 35% COMPLETE FOR L2
PROCESSING IS APPROXIMATELY 54% COMPLETE FOR L2
PROCESSING IS APPROXIMATELY 73% COMPLETE FOR L2
PROCESSING IS APPROXIMATELY 93% COMPLETE FOR L2
PROCESSING COMPLETED FOR L2
           6958 DUP REM L1 L2 (88 DUPLICATES REMOVED)
L4
=> s 14 and (esterif? or transesterif?)
          514 S L4
L5
          6444 S L4
L6
        121695 ESTERIF?
         19336 TRANSESTERIF?
           745 (L5 OR L6) AND (ESTERIF? OR TRANSESTERIF?)
L7
=> s 17 and alcohol
        214408 ALCOHOL
        147720 ALCOHOLS
        335134 ALCOHOL
                  (ALCOHOL OR ALCOHOLS)
        538090 ALC
        179251 ALCS
        629729 ALC
                 (ALC OR ALCS)
        748695 ALCOHOL
                 (ALCOHOL OR ALC)
           244 L7 AND ALCOHOL
L8
=> s 14 and (process or prepar? or make or made or synthes?)
           514 S L4
L9
          6444 S L4
L10
       1958444 PROCESS
       1299692 PROCESSES
       2911574 PROCESS
```

```
(PROCESS OR PROCESSES)
      1480947 PREPAR?
       110746 PREP
         1954 PREPS
       112505 PREP
                 (PREP OR PREPS)
      1874342 PREPD
           21 PREPDS
      1874357 PREPD
                 (PREPD OR PREPDS)
        97386 PREPG
           12 PREPGS
         97397 PREPG
                 (PREPG OR PREPGS)
      2490259 PREPN
       195592 PREPNS
      2638614 PREPN
                 (PREPN OR PREPNS)
       4367118 PREPAR?
                 (PREPAR? OR PREP OR PREPD OR PREPG OR PREPN)
       192678 MAKE
        148039 MAKES
        331508 MAKE
                 (MAKE OR MAKES)
       1103437 MADE
            24 MADES
       1103457 MADE
                 (MADE OR MADES)
       1381870 SYNTHES?
         4198 (L9 OR L10) AND (PROCESS OR PREPAR? OR MAKE OR MADE OR SYNTHES?)
L11
=> s l11 and (esterif? or transesterif?)
        121695 ESTERIF?
         19336 TRANSESTERIF?
           594 L11 AND (ESTERIF? OR TRANSESTERIF?)
L12
=> s l12 and alcohol
        214408 ALCOHOL
        147720 ALCOHOLS
        335134 ALCOHOL
                  (ALCOHOL OR ALCOHOLS)
        538090 ALC
        179251 ALCS
        629729 ALC
                  (ALC OR ALCS)
        748695 ALCOHOL
                  (ALCOHOL OR ALC)
           193 L12 AND ALCOHOL
L13
=> s 113 and inhibitor
        447426 INHIBITOR
        466871 INHIBITORS
        720148 INHIBITOR
                  (INHIBITOR OR INHIBITORS)
             52 L13 AND INHIBITOR
L14
=> s 113 and polymerization inhibitor
         309594 POLYMERIZATION
          3945 POLYMERIZATIONS
        310136 POLYMERIZATION
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302253 POLYMN 7687 POLYMNS

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303220 POLYMN
                 (POLYMN OR POLYMNS)
        418088 POLYMERIZATION
                 (POLYMERIZATION OR POLYMN)
        447426 INHIBITOR
        466871 INHIBITORS
        720148 INHIBITOR
                 (INHIBITOR OR INHIBITORS)
          7284 POLYMERIZATION INHIBITOR
                 (POLYMERIZATION (W) INHIBITOR)
            50 L13 AND POLYMERIZATION INHIBITOR
1.15
=> d dup rem 114 115
L15 IS NOT VALID HERE
For an explanation, enter "HELP DISPLAY".
=> dup rem 114 115
PROCESSING COMPLETED FOR L14
PROCESSING COMPLETED FOR L15
             52 DUP REM L14 L15 (50 DUPLICATES REMOVED)
L16
=> d 116 ibib hitstr abs 1-52
L16 ANSWER 1 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 1
                        2004:139077 CAPLUS
ACCESSION NUMBER:
                         140:181967
DOCUMENT NUMBER:
                         Process for manufacturing (meth)
TITLE:
                         acrylic esters
                         Hayashi, Katsunori; Kobayashi, Akihiro; Arashima,
INVENTOR(S):
                         Katsuyasu
                         Hitachi Chemical Co., Ltd., Japan
PATENT ASSIGNEE(S):
                         Jpn. Kokai Tokkyo Koho, 7 pp.
SOURCE:
                         CODEN: JKXXAF
                         Patent
DOCUMENT TYPE:
                         Japanese
LANGUAGE:
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
                 KIND DATE APPLICATION NO. DATE
     PATENT NO.
     _____
     JP 2004051546 A2
                                        JP 2002-211205 20020719
                            20040219
                                        JP 2002-211205 20020719
PRIORITY APPLN. INFO.:
     In the process for manufacturing acrylic ester or methacrylic ester by
     reacting an alc. with acrylic acid or methacrylic acid in a
     solvent in the presence of an acid catalyst, the reaction mixture is
     neutralized (upon completion of the esterification), washed with water, and a basic lithium salt (at 0.01 to 5 weight% relative to the ester
     produced) is added to the neutralized liquid, the reaction solvent is distilled
     off while mol. oxygen is introduced, and the insol. matter is removed by
     filtration. Thus, tris(2-hydroxyethyl) isocyanurate acrylate (I), showing
     APHA 40 and good solubility in MeOH and toluene, was prepared in 95%
     yield, vs. I (prepared by a reference process) showing APHA
     40 with good solubility in methanol and low solubility in toluene.
L16 ANSWER 2 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 2
ACCESSION NUMBER:
                         2004:159023 CAPLUS
DOCUMENT NUMBER:
                         140:199903
```

(POLYMERIZATION OR POLYMERIZATIONS)

```
Esterification process and
TITLE:
                          catalysts for the production of (meth) acrylate esters
                          from (meth) acrylic acid
                          and alcohols
                          Beste, York Alexander; Stein, Bernd
INVENTOR(S):
                        BASF A.-G., Germany
PATENT ASSIGNEE(S):
                         Ger. Offen., 11 pp.
SOURCE:
                         CODEN: GWXXBX
DOCUMENT TYPE:
                         Patent
LANGUAGE:
                          German
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
     PATENT NO. KIND DATE APPLICATION NO. DATE
     DE 10317435 A1 20040226 DE 2003-10317435 20030415
RITY APPLN. INFO.: DE 2003-10317435 20030415
PRIORITY APPLN. INFO.:
     (meth)acrylate esters (e.g., Bu acrylate) are prepared by the
     heterogeneous-catalyzed (e.g., Amberlyst 15) conversion of (meth
     )acrylic acids with at least one alc. in at
     least one reactor, in which: (A) the water content in the bottom of the
     azeotropic distillation column is >0.15 ppm; and/or (B) the content of the (
     meth) acrylic acid in the sump of the azeotrope
     column is ≤60%; and/or (C) the heterogeneous catalyst before the
     reaction is contacted with a polymerization inhibitor-containing
     alc. solution (e.g., 1-butanol and phenothiazine).
L16 ANSWER 3 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 3
ACCESSION NUMBER: 2003:610404 CAPLUS
                         139:149334
DOCUMENT NUMBER:
                         Process for preparation of (
TITLE:
                         meth) acrylic acid esters
INVENTOR(S):
                         Yada, Shuhei; Goriki, Masayasu; Nakamura, Mitsuo
                       Mitsubishi Chemical Corporation, Japan
PATENT ASSIGNEE(S):
                         PCT Int. Appl., 11 pp.
SOURCE:
                          CODEN: PIXXD2
                          Patent
DOCUMENT TYPE:
                          Japanese
LANGUAGE:
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
                  KIND DATE APPLICATION NO. DATE
     PATENT NO.
     _____
                                            ______
     WO 2003064370 A1 20030807 WO 2003-JP594 20030123
         W: AE, AG, AL, AU, BA, BB, BR, BZ, CA, CN, CO, CR, CU, DM, DZ, EC,
             GD, GE, HR, HU, ID, IL, IN, IS, KP, KR, LC, LK, LR, LT, LV, MA, MG, MK, MN, MX, NO, NZ, OM, PH, PL, RO, SC, SG, TN, TT, UA, US,
         UZ, VC, VN, YU, ZA, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC,
             NL, PT, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
                                           JP 2003-13454
     JP 2003292472
                     A2 20031015
                                                               20030122
                                        JP 2002-21374 A 20020130
PRIORITY APPLN. INFO.:
     This invention pertains to a method for producing (meth)
     acrylic acid esters which comprises feeding (
     meth) acrylic acid and an alc. to a
     reactor, wherein a polymerization inhibitor is dissolved in the
     reaction mixture (Meth) acrylic acid and (
     meth) acrylic acid esters are prevented from
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10/657,157
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polymerizing on the member present in the gas-phase part of the reactor, and a continuous operation can be conducted over long. For example, acrylic acid was reacted with BuOH in the presence of p-toluenesulfonic acid and hydroquinone to give Bu acrylate.

REFERENCE COUNT: 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L16 ANSWER 4 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 4

ACCESSION NUMBER:

2003:396830 CAPLUS

DOCUMENT NUMBER:

138:385915

TITLE:

Method for producing (meth)acrylic acid esters of polyhydric alcohols

INVENTOR(S):

Martin, Friedrich-Georg; Wartini, Alexander; Dernbach,

Matthias; Schroeder, Juergen; Sirch, Tilman

PATENT ASSIGNEE(S):

BASF Aktiengesellschaft, Germany PCT Int. Appl., 41 pp.

SOURCE:

CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT: 6

PATENT INFORMATION:

```
KIND DATE APPLICATION NO. DATE
                  KIND DATE
    PATENT NO.
                                              ______
     _____
                                             WO 2002-EP12491 20021108
    WO 2003042151 A1 20030522
WO 2003042151 C1 20040624
                              20030522
    WO 2003042151
            PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD,
              RU, TJ, TM
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR,
              NE, SN, TD, TG
                                               DE 2001-10156116 20011115
                       A1 20030626
     DE 10156116
                                           DE 2001-10156116 A 20011115
PRIORITY APPLN. INFO.:
                           MARPAT 138:385915
OTHER SOURCE(S):
     (Meth) acrylic acid esters of polyhydric
     alcs. are manufactured by reacting (meth) acrylic
     acid and the corresponding polyhydric alcs. in the
     presence of ≥1 acid catalyst and, optionally, ≥1 polymerization
     inhibitor and a solvent, whereby the polyhydric alc.
     contains <500 ppm HCHO. Thus, trimethylolpropane containing 282 ppm
     acetal-bound HCHO was esterified with acrylic acid in
     cyclohexane mixture containing p-MeOC6H4OH, H3PO2, CuCl2 and H2SO4 to give
```

for similar product **prepared** by use of trimethylolpropane containing
1400 ppm of acetal-bound HCHO.

REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS
RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L16 ANSWER 5 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 5

138:188258

product having d. 1.1041 g/cm3 and dynamic viscosity 85 mPa·s (23°), vs. d. 1.1153 g/cm3 and dynamic viscosity 246 mPa·s

ACCESSION NUMBER: 2003:166982 CAPLUS

DOCUMENT NUMBER:

Preparation of 2-phenylethyl (meth)acrylate

INVENTOR(S):

TITLE:

Doi, Junichi; Sonobe, Hiroshi; Matsumoto, Satoshi

PATENT ASSIGNEE(S):

Mitsubishi Rayon Co., Ltd., Japan

SOURCE:

Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent Japanese

LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE PATENT NO. KIND DATE \_\_\_\_\_ \_\_\_\_\_ JP 2003064024 A2 JP 2001-254623 20010824 20030305 JP 2001-254623 20010824 PRIORITY APPLN. INFO.:

The compound is prepared by esterification of (

meth) acrylic acid with 2-phenylethyl

alc. in the presence of 0.05-0.5 mol equivalent (based on 1 mol alc.) of acid catalysts. Methacrylic acid was esterified
with 2-phenylethyl alc. in the presence of p-MeC6H4SO3H, N,N'-di-2-naphthyl-p-phenylenediamine, and p-methoxyphenol at 72-83° for 10 h, washed with alkalies, and distilled to give 95% 2-phenylethyl methacrylate with 99.7% purity.

L16 ANSWER 6 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN

ACCESSION NUMBER: 2003:35382 CAPLUS

DOCUMENT NUMBER:

138:73685

TITLE:

Process for preparing alkyl

(meth) acrylates

INVENTOR(S):

Venter, Jeremia Jesaja; Mirabelli, Mario Giuseppe

Luciano

PATENT ASSIGNEE(S):

Rohm and Haas Company, USA

SOURCE:

U.S., 7 pp., Cont.-in-part of U.S. Ser. No. 192,675,

abandoned. CODEN: USXXAM

DOCUMENT TYPE:

Patent

LANGUAGE:

English

5

FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE PATENT NO. US 6506930 B1 20030114 US 1999-417400 19991013
RITY APPLN. INFO.: US 1997-66939P P 19971117
US 1998-192675 B2 19981116 PRIORITY APPLN. INFO.:

The process provides for the synthesis of alkyl (meth) acrylates, hydrolysis of process impurities into starting materials and separation of starting materials and reaction products in one reactor. A typical process comprises: (A) charging a reactor with a C1-4 alc.; a (meth) acrylic acid; a strong acid catalyst selected from the group consisting of sulfuric acid, alkylsulfonic acid and polymer supported alkylsulfonic acid; at least one inhibitor selected from the group consisting of 2,2,6,6-tetramethyl-1-piperidinyloxy, 4-hydroxy-2,2,6,6-tetramethyl-1piperidinyloxy, 4-methacryloyloxy-2,2,6,6-tetramethyl piperidinyl free radical and 4-hydroxy-2,2,6,6-tetramethyl N-hydroxypiperidine; and at least 5% water to form a reaction mixture; (B) reacting the reaction mixture to form a C1-4 alkyl (meth) acrylate and process impurities, wherein the process impurities are hydrolyzed in said reactor; and (C) separating the C1-4 alkyl (meth) acrylate and water formed during the reaction from the reaction mixture

REFERENCE COUNT:

THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L16 ANSWER 7 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 6

ACCESSION NUMBER: 2002:845303 CAPLUS

DOCUMENT NUMBER:

137:338384

TITLE:

Esterification process for the production of (meth)acrylate esters

INVENTOR(S):

Nestler, Gerhard; Geisendoerfer, Matthias

PATENT ASSIGNEE(S): BASF AG, Germany

Ger. Offen., 12 pp.

SOURCE:

CODEN: GWXXBX

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

DE 10154714 A1 20021107 DE 2001-10154714 20011109

DE 10246869 A1 20030327 DE 2002-10246869 20021008

RITY APPLN. INFO.: DE 2001-10154714 A1 20011109 PRIORITY APPLN. INFO.: The production of (meth) acrylate esters (e.g., 2-ethylhexyl acrylate) by is

achieved by acid-catalyzed esterification of (meth) acrylic acids (e.g., acrylic acid) with the appropriate alc. (e.g., 2-ethylhexanol) in a homogeneous liquid phase in the presence of a polymerization inhibitor (e.g., phenothiazine) and/or an inhibitor mixture, one accomplishes the esterification (stage 1) in a reaction zone, which is equipped with at least one distillation unit, over which one separates the reaction water as well as olefins, alc., acetic acid esters and propionic acid ester, formed with the esterification, condensed and in an aqueous and an organic phase are separated, the discharge from the reaction zone from stage 1 is lead into a catalyst separation stage (stage 2) and into a esterification -catalyzed bottoms product and the (meth)acrylate esters head product is

separated, from this (meth) acrylate ester-containing head product in a following

stage the remaining (meth)acrylate ester-containing stream/current is separated into a light-boiling fraction (stage 4) and the (meth)acrylate ester essentially freed of acetic acid ester and output alc. recycled, from the released acetic acid ester and output alc. the (meth)acrylate ester made from stage 4 in a pure distillation (stage 6) separates from the high-boiling solvents and the high-boiling solvent-containing stream is subjected to a thermal and/or catalytic treatment.

L16 ANSWER 8 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 7

ACCESSION NUMBER:

2002:403632 CAPLUS

DOCUMENT NUMBER:

136:402195

TITLE:

SOURCE:

Transesterification process for

the production of higher alkyl (meth)acrylate esters

from lower-alkyl (meth)acrylate esters

INVENTOR(S):

Nestler, Gerhard; Rauh, Ulrich; Schroeder, Juergen BASF AG, Germany

Ger. Offen., 12 pp. CODEN: GWXXBX

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT ASSIGNEE(S):

PATENT NO. KIND DATE

APPLICATION NO. DATE

```
DE 10127941 A1 20020529 DE 2001-10127941 20010608 WO 2002100815 A1 20021219 WO 2002-EP5821 20020528
                                        _____
        W: CN, JP, US
        RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
            PT, SE, TR
                          20040324 EP 2002-747325
                                                         20020528
                    A1
    EP 1399409
        R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
            IE, FI, CY, TR
                                      DE 2001-10127941 A 20010608
PRIORITY APPLN. INFO.:
                                      WO 2002-EP5821
                                                     W 20020528
OTHER SOURCE(S):
                       MARPAT 136:402195
    (meth)acrylate esters [e.g., 2-(dimethylamino)ethyl acrylate] are
    prepared in high yield and selectivity by the
    transesterification of lower-alkyl (meth)acrylate esters (e.g., Bu
    acrylate) with a higher alc. [e.g., 2-(dimethylamino)ethanol] in
    the presence of a polymerization inhibitor (e.g., phenothiazine and
    hydroquinone monomethyl ether) and a transesterification
    catalyst (e.g., tetra-Bu titanate) or a catalyst mixture, and where one
    separates the free byproduct lower alkanol (e.g., 1-butanol) and at least
    partly supplies the production of the lower-alkyl (meth) acrylate by
    acidification of the process waste water and
    esterification of the residual (meth) acrylic
    acid (e.g., acrylic acid) with the byproduct alkanol.
L16 ANSWER 9 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 8
ACCESSION NUMBER: 2002:462390 CAPLUS
                        137:33676
DOCUMENT NUMBER:
                        Procedure for manufacture of (meth)
TITLE:
                        acrylic acid esters
                        Martin, Friedrich-Georg; Nestler, Gerhard; Schroeder,
INVENTOR(S):
                        Juergen
                        BASF AG, Germany
PATENT ASSIGNEE(S):
                        Ger. Offen., 4 pp.
SOURCE:
                        CODEN: GWXXBX
DOCUMENT TYPE:
                        Patent
                        German
LANGUAGE:
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
                                  APPLICATION NO. DATE
     PATENT NO. KIND DATE
                                         _____
     ______
                                       DE 2000-10063176 20001218
     DE 10063176 A1
                           20020620
                                         WO 2001-EP14903 20011217
                    A1 20020627
     WO 2002050015
         W: US
         RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
             PT, SE, TR
                                        EP 2001-984858
                                                          20011217
                          20030924
                     A1
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, FI, CY, TR
     US 2004030179 A1 20040212
                                       US 2003-433614
                                                          20030617
                                       DE 2000-10063176 A 20001218
PRIORITY APPLN. INFO.:
                                       WO 2001-EP14903 W 20011217
     (meth) acrylic acid esters (of mol. weight >200)
AR
     are obtained by esterification of (meth)
     acrylic acid with alcs. in the presence of
     \geq1 acid catalyst, \geq1 polymerization inhibitor, and an
     organic solvent, which forms an azeotrope with water, whereby the the mixture
     is heated to the b.p. in an apparatus with a distillation unit, column and
condenser,
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the azeotrope is distilled off and the organic solvent is recirculated to the column at least partially contacting a copper-containing material, such as distillation column packings or separation efficient fittings. Thus, acrylic acid

2380, tripropylene glycol 2880, cyclohexane 2300, p-toluenesulfonic acid 120, 50% phosphinic acid 9.4, and hydroquinone monomethylether 4.7 parts were mixed in a 10-L-reactor with a double layer heating and distillation

(5 + 60 cm). The reaction water formed was distilled off as azeotrope with cyclohexane, whereby after after condensation the organic phase formed was separated and recirculated to the column, which was filled with copper Raschig-rings at the top and glass rings under it. Within 8 h, 546 parts of water was separated, so that a 97% yield of esterification took place and no polymerization was observed in the column.

L16 ANSWER 10 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 9

ACCESSION NUMBER:

2002:462389 CAPLUS

DOCUMENT NUMBER:

137:47591

TITLE:

Procedure for manufacture of higher (meth)

acrylic acid esters

INVENTOR(S):

Martin, Friedrich-Georg; Nestler, Gerhard; Schroeder,

Juergen

PATENT ASSIGNEE(S):

BASF AG, Germany Ger. Offen., 6 pp.

SOURCE:

CODEN: GWXXBX

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT: 6

PATENT INFORMATION:

	KIND DATE	APPLICATION NO. DATE
DE 10063175 WO 2002055472 W: US	A1 20020620 A1 20020718	DE 2000-10063175 20001218
RW: AT, BE,	TR	ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
EP 1345886 R: AT, BE,	Al 20030924 CH, DE, DK, ES,	EP 2001-984833 20011213 FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
TE, F1, WO 2002050014 W: US	CY, TR Al 20020627	WO 2001-EP14902 20011217
RW: AT, BE,	TR	ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
EP 1343748 R: AT, BE,	A1 20030917 CH, DE, DK, ES,	EP 2001-994797 20011217 FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
US 2004019235	CY, TR A1 20040129 A1 20040205	US 2003-433612 20030617 US 2003-450438 20030618
PRIORITY APPLN. INFO	.:	DE 2000-10063175 A 20001218 DE 2001-10152680 A 20011019
		WO 2001-EP14636 W 20011213 WO 2001-EP14902 W 20011217

Higher (meth) acrylic acid esters (of mol. AΒ weight >200) are obtained by esterification of (meth) acrylic acid with higher alcs. in the presence of ≥1 acid catalyst, ≥1 polymerization inhibitor, and an organic solvent, which forms an azeotrope with water, whereby the the mixture is heated to the b.p. in an apparatus with a distillation unit, column and condenser

and the reactor content is circulated via an external evaporator, preferably an self-circulating evaporator. Thus, (meth) acrylic acid 172, methoxyphenol 9.2, phenothiazine 0.3, 50% phosphinic acid 23.8, 65% p-toluenesulfonic acid 497, and cyclohexane 1030 g were mixed in a 10-L-reactor with an external evaporator, distillation column, and condenser. A circulation evaporator (a heat exchanger comprising a nest of boiler tubes filled with oil) was used. After evaporating the water from the reaction mixture, 344 g (meth) acrylic acid and 6000 g Me polyethylene glycol were added to give after 330 min 207 g aqueous phase and 6890 g crude ester. The procedure permits the preparation of higher (meth) acrylic acid esters without use of copper salts as polymerization inhibitors.

L16 ANSWER 11 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 10

ACCESSION NUMBER: 2001:481835 CAPLUS

DOCUMENT NUMBER: 135:61734

TITLE: Manufacture of (meth) acrylate esters with recycling

polymerization inhibitor-containing residues

INVENTOR(S): Nakahara, Osamu; Kamioka, Masatoshi

PATENT ASSIGNEE(S): Nippon Shokubai Kagaku Kogyo Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2001181233	A2	20010703	JP 2000-311413	20001012
US 6649787	В1	20031118	US 2000-684396	20001010
RITY APPLN. INFO.	:		JP 1999-289345 A	19991012

PRIORITY APPLN. INFO.: JP 1999-289345 A 1999-101
AB (meth)acrylate esters are manufactured by **esterification** of (

meth) acrylic acid with C1-4 aliphatic

alcs. in the presence of acid catalysts and distillation of the reaction
 mixts. using an acid separation column., a low boiling substance-separation
column,

and a rectifying column with returning ..... from the rectifying column to the <code>esterification</code> and/or the separation/purification <code>process</code>. Thus, (a) acrylic acid containing phenothiazine (I), (b) BuOH, and (c) I-containing high boiling residue collected from the rectifying column were passed through a cation exchanger-packed rector and the reaction mixture was distilled with using fresh I as a polymerization <code>inhibitor</code> in the acid separation and rectifying <code>processes</code>. Hydroquinone monomethyl ether was also used as a polymerization <code>inhibitor</code> in another rectifying column. and Bu acrylate was collected. The polymerization <code>inhibitors</code> were recovered and reused in distillation of the acrylate ester.

L16 ANSWER 12 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 11

ACCESSION NUMBER: 2001:150618 CAPLUS

DOCUMENT NUMBER: 134:193858

TITLE: Esterification process for the

production of (meth) acrylate esters with the

azeotropic distillation of process water

INVENTOR(S): Leube, Hartmann F.; Leidinger, Kurt; Geisendoerfer,

Matthias; Beck, Erich

PATENT ASSIGNEE(S): BASF A.-G., Germany

SOURCE: Ger. Offen., 6 pp.

CODEN: GWXXBX

DOCUMENT TYPE: Patent

LANGUAGE: GEFAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

DE 19941136 A1 20010301 DE 1999-19941136 19990830

German

EP 1081125 A1 20010307 EP 1081125 B1 20030611

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,

IE, SI, LT, LV, FI, RO

AT 242759 E 20030615 AT 2000-117813 20000818 ES 2200763 T3 20040316 ES 2000-117813 20000818 US 6437172 B1 20020820 US 2000-644824 20000824 PRIORITY APPLN. INFO.: DE 1999-19941136 A 19990830

AB (meth)acrylate esters (e.g., the methacrylate ester of propoxylated trimethylolpropane) are prepared in high yield and selectivity by the esterification of a high-boiling mono- or polyhydric

alc. (e.g., propoxylated trimethylolpropane) with (meth)

acrylic acid in the presence of an
esterification catalyst (e.g., p-toluenesulfonic acid), a polymerization
inhibitor (e.g., hydroquinone monomethyl ether), and an azeotropic

distillation agent, such as (cyclo)aliphatic hydrocarbons (e.g., cyclohexane),

for

the removal of **process** water, which azeotropic distillation agent is dosed into the reaction mixture during the **esterification** reaction for both the removal of **process** water and so as to control the boiling temperature of the reaction.

L16 ANSWER 13 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 12

ACCESSION NUMBER:

2001:114630 CAPLUS

DOCUMENT NUMBER:

134:163471

TITLE:

Production of esters of (meth)

acrylic acid and polyhydric
alcohols and device therefor

INVENTOR(S):

Roessler, Harald; Fies, Matthias; Gutsche, Bernhard;

WO 2000-EP7496 W 20000803

EP 2000-117813 20000818

Stalberg, Theo

PATENT ASSIGNEE(S):

Cognis Deutschland G.m.b.H., Germany

SOURCE: Ger. Offen., 6 pp.

CODEN: GWXXBX

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND DAT	E A	APPLICATION NO.	DATE
DE 19937911	A1 200	 10215	E 1999-19937911	19990811
WO 2001012315			O 2000-EP7496	20000803
W: KR, US RW: AT, BE,	CH, CY, DE	, DK, ES, FI,	FR, GB, GR, IE	, IT, LU, MC, NL,
PT, SE EP 1204472	A1 200	20515 F	EP 2000-958367	20000803
EP 1204472 R: AT, BE,		30129 , ES, FR, GB,	GR, IT, LI, LU	, NL, SE, MC, PT,
IE, FI, ES 2191637	CY		ES 2000-958367	20000803
PRIORITY APPLN. INFO	. :	DE 3	1999-19937911 A	19990811

AB Esters based on (meth)acrylic acid and

polyhydric alcs. are produced in a reactor whereby the liquid reaction mixture contains polymerization inhibitors and whereby one at least partly separates the produced water of reaction in the vapor state, which is characterized by the fact that a part of the gas/vapor phase of the reaction mixture is removed from the reactor, then partially condensed in a dephlegmator and then fed to an ascending gas/steam mixture as well as the condensate coming from the head of the dephlegmator, with a polymerization inhibitor being added, and with provisions being made

for recycling to the reactor. The **process** permits the use of a smaller excess of unsatd. acid, decreases the chance of polymerization,

shortens

the reaction time, and decreases the organic wastewater content.

L16 ANSWER 14 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 13

ACCESSION NUMBER:

2000:600284 CAPLUS

DOCUMENT NUMBER:

133:193609

TITLE:

Radical polymerization **inhibitors** containing phenothiazine adducts and manufacture of

(meth)acrylate esters using them

INVENTOR(S):

Sato, Hiroyuki; Fukumura, Takanori; Ohizumi, Fumitaka

Chisso Corp., Japan

PATENT ASSIGNEE(S): SOURCE:

Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent Japanese

LANGUAGE:

· 1

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 2000234002 A2 20000829 JP 1999-36716 19990216

PRIORITY APPLN. INFO.: JP 1999-36716 19990216

AB The inhibitors comprise adducts of phenothiazine with S-containing

acidic compds. and are used for **preparation** of (meth)acrylate esters from (meth)acrylic acid or (meth)acryloyl chloride and hydroxy compds. Thus, a reaction of methacrylic acid with 1-(2-hydroxyethyl)-2-pyrrolidone at 130° for 4 h in the presence of H2SO4 and phenothiazine gave 2-(2-pyrrolidonyl)ethyl methacrylate.

L16 ANSWER 15 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 14

ACCESSION NUMBER:

2000:412204 CAPLUS

DOCUMENT NUMBER:

133:43958

TITLE:

SOURCE:

Preparation of high-purity (meth)

acrylic acid esters

INVENTOR(S):

Yoshida, Koichi; Tokuda, Masanori; Sonobe, Hiroshi;

Ohkita, Motomu

PATENT ASSIGNEE(S):

Mitsubishi Rayon Co., Ltd., Japan

Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 2000169429 A2 20000620 JP 1998-351392 19981210

PRIORITY APPLN. INFO.: JP 1998-351392 19981210

OTHER SOURCE(S):

MARPAT 133:43958

GI

The title compds. are **prepared** by **transesterification** of Me (meth)acrylate with C3-20 **alcs**. or phenols in the presence of supported transition metal compound catalysts and I [R1-R4 = alkyl; R5 = H, OH, OR, OCOR, NHCOR, O[(C2H4O)n + (C3H6O)m]H; R6 = H, or R5R6 = :O; R = (un)substituted C1-18 alkyl, alkenyl, aryl; m, n = 0-10, and m = n ≠ 0]. Thus, stirring MMA with allyl **alc**. in the presence of silica/alumina-supported Ti catalysts and I (R1-R4 = Me; R5 = OH; R6 = H) gave allyl methacrylate with 99.5% purity.

L16 ANSWER 16 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 15

ACCESSION NUMBER:

2000:374769 CAPLUS

DOCUMENT NUMBER:

133:8284

TITLE:

Polyoxyalkylene (meth)acrylic

acid esters and their polymers, their

manufacture, and use of the polymers as cement

dispersants

INVENTOR (S):

Hirata, Takeshi; Yuasa, Tsutomu

PATENT ASSIGNEE(S):

Nippon Shokubai Kagaku Kogyo Co., Ltd., Japan

Jpn. Kokai Tokkyo Koho, 19 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

SOURCE:

Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

APPLICATION NO. DATE KIND DATE PATENT NO. \_\_\_\_\_ \_\_\_\_\_\_ JP 1998-328686 19981118 20000606 A2 JP 2000154247 JP 1998-328686 19981118 PRIORITY APPLN. INFO.: CH2:CR3CO2(R2O)nR1 (I; R1 = C1-30 hydrocarbon; R2O = C2-18 oxyalkylene; R2O may be block or random copolymers; R3 = H, Me; n = average oxyalkylene groups = 0-300) containing  $\leq 2.0$  area% (in liquid chromatog.) polymerized products is claimed. I is manufactured by esterification reaction of R1(R2O)nH and (meth)acrylic acid in presence of a polymerization inhibitor in a dehydration solvent, and addition of ≤1000 ppm (based on the total amount of raw material alc. and acid) water-soluble polymerization inhibitor during removal of the dehydration solvent from the reaction solution, after finishing the esterification. Polymers containing I, their manufacture, and their use in cement dispersants are also claimed. Cement compns. containing the polymers showed high slump maintaining properties.

L16 ANSWER 17 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 16

ACCESSION NUMBER: 2000:277696 CAPLUS

DOCUMENT NUMBER: 132:279644

TITLE: Process for the preparation and

extraction of (meth)acrylate esters of mono- and polyhydric alcohols using dense-phase carbon

dioxide

INVENTOR(S): PATENT ASSIGNEE(S):

Tweedy, Harrel E. UCB, S.A., Belg.

SOURCE:

Eur. Pat. Appl., 18 pp.

CODEN: EPXXDW

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

APPLICATION NO. DATE KIND DATE PATENT NO. \_\_\_\_\_ \_\_\_\_\_ EP 995738 A1 20000426 EP 1999-120221 19991011

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,

IE, SI, LT, LV, FI, RO

JP 1999-296617 19991019 A2 20000509 JP 2000128829 US 1998-174412 A 19981019

PRIORITY APPLN. INFO.: The present invention relates to a process for preparing acrylate esters, methacrylate esters, polyester acrylates (e.g., trimethylolpropane triacrylate) and/or polyester methacrylates by reacting acrylic and/or methacrylic acid with a monohydroxy compound and/or a polyhydroxy compound (e.g., trimethylolpropane) in a reaction vessel, in the presence of a catalyst (e.g., Amberlyst A15), a polymerization inhibitor (e.g., phenothiazine), and dense-phase carbon dioxide. The dense phase carbon dioxide is in either a sub-critical, near-critical, critical, or supercrit.

phys. state. The advantages of using dense-phase carbon dioxide as the solvent include the preparation of (meth) acrylate esters and polyester (meth) acrylates in an environmentally friendly manner by reducing or eliminating the need for hydrocarbon solvents, reducing or eliminating the problems associated with handling hydrocarbon solvents, simplifying recovery of the products, and recovery and recycle of reactants as compared to the conventional esterification synthesis. In addition, these products have improved color and

purity.

REFERENCE COUNT: RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L16 ANSWER 18 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 17

ACCESSION NUMBER:

2000:240714 CAPLUS

DOCUMENT NUMBER:

132:279640

TITLE:

Preparation of (meth) acrylate esters and polyester (meth) acrylates in the presence of a

THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS

catalyst and inhibitor under microwave

heating

TNVENTOR(S):

Tweedy, Harrel E. UCB, S.A., Belg.

PATENT ASSIGNEE(S): SOURCE:

Eur. Pat. Appl., 15 pp.

CODEN: EPXXDW

DOCUMENT TYPE:

Patent English

LANGUAGE: MILY ACC. NUM. COUNT:

ENT INFORMATION:

APPLICATION NO. DATE KIND DATE ENT NO. \_\_\_\_\_ \_\_\_\_\_ A1 20000412 EP 1999-119616 19991004

I, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,

10/657,157 IE, SI, LT, LV, FI, RO US 1998-168991 20010116 US 6175037 В1 JP 1999-289458 20000425 A2 JP 2000119216 US 1998-168991 A 19981009 PRIORITY APPLN. INFO.: Acrylic or methacrylic acid is reacted with a monohydroxy- or polyhydroxy-containing compound in the presence of a catalyst and polymerization inhibitor using microwave energy as the heating source, which can be applied to the reaction vessel continuously or intermittently by pulsing. Preferably the monohydroxy-containing compds. are C1-18 linear or branched aliphatic, cycloaliph. and aromatic compds., such as methanol and cyclohexano; and the polyhydroxy-containing compds. are C2-36 polyols having .apprx.2-10 hydroxy groups, such as cyclohexandiol, butanediol and tripropylene glycol. Advantages of using microwave energy include higher temps. coupled with shorter residence times, reduced production costs, increased capacity, lower energy costs, effective use of raw materials, and solventless processing which is environmentally friendly. Thus, a mixture of 7.3 g methacrylic acid, 5.0 g 1,6-hexanediol, 0.1 g phenothiazine, and 0.6 g methanesulfonic acid (70% aqueous solution) was polymerized in a microwave oven at 20% power (.apprx.200 W) for 30 s followed by 20 s shaking for 2 cycles, and 18 cycles of 20-s heating/20-s agitation. After a total of 7 min cumulative microwave exposure the sample contained (GC, area % values) 74.2% 1,6-hexanediol dimethacrylate, 22.2% 1,6-hexanediol monomethacrylate, and 1.3% 1,6-hexanediol along with some other minor peaks. THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT L16 ANSWER 19 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 18 2000:209660 CAPLUS ACCESSION NUMBER: 132:237568 DOCUMENT NUMBER: Method for production of polyoxyalkylene acrylates TITLE: Hirata, Tsuyoshi; Yuasa, Tsutomu INVENTOR(S): Nippon Shokubai Co., Ltd., Japan; Nippon Catalytic PATENT ASSIGNEE(S): Chem. Ind. Eur. Pat. Appl., 36 pp. SOURCE: CODEN: EPXXDW Patent DOCUMENT TYPE: English LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION:

	PAT	ENT	NO.		KII	ND	DATE			AF	PLI	CAT	ION	NC	).	DATE			
	ΕP	9891	.08		A:	2	2000	0329		EF	19	99-	118	597	,	1999	0921		
	ΕP	9891	.08		A.	3	2001	0124											
	ΕP	9891	.08		<b>B</b> :		2003												
		R:	AT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	IT	', L	I,	LU,	NL,	SE,	MC,	PT,
			ΙE,	SI,	LT,	LV,	FI,	RO											
	US	6265	495		В	1	2001	0724		US	3 19	99-	399	491	-	1999	0920		
	KR	2000	0234	20	Α		2000	0425		KF	19	99-	411	17		1999	0922		
	JP	2000	1598	82	A	2	2000	0613		JI	19	99-	269	435	5	1999	0922		
	JP	3094	024		B	2	2000	1003											
	CN	1256	265		Α		2000	0614		CI	1 19	999-	125	143	}	1999	0922		
	CN	1134	406		В		2004	0114											
	JΡ	2000	2121	28	A	2	2000	0802		JI	19	999-	311	664	Ŀ	1999	1101		
	JΡ	3390	382		B	2	2003	0324											
PRIOR	RITY	APF	LN.	INFO	. :				Ċ	JP 19	98-	-268	122		Α	1998	0922		
									į	JP 19	98-	-328	687		Α	1998	1118		

A method for the production of an esterified product is provided AB which affords an esterified product with high quality by repressing, during the esterification reaction of an alc . with (meth) acrylic acid, the occurrence of impurities, particularly gel, to be formed owing to the polymerization of the alc. and the (meth)acrylic acid as raw materials, the esterified product consequently formed, or the mixture thereof. Specifically, a method for the production of an esterified product of this invention comprises esterifying an alc. represented by the following formula (1): R10(R20) nH wherein R1 represents a hydrocarbon group of 1 to 30 carbon atoms, R20 represents an oxyalkylene group of 2 to 18 carbon atoms, providing that the repeating units, R2O, may be the same or different and that when the R20's are in the form of a mixture of two or more species, the repeating units, R2O, may be added either in a block form or in a random form, and n represents an average addition mol number of oxyalkylene groups and is in the range

of 0 to 300, with (meth)acrylic acid in a dehydrating solvent in the presence of an acid catalyst and a polymerization inhibitor, wherein a reaction temperature during the esterification reaction is not higher than 130°C and a circulation speed of the solvent during the esterification reaction is not less than 0.5 cycle/h.

L16 ANSWER 20 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 19

ACCESSION NUMBER:

2000:909652 CAPLUS

DOCUMENT NUMBER: TITLE:

134:72023 Process for the manufacture of (meth

)acrylic acid esters

INVENTOR(S):

Paulus, Wolfgang; Reich, Wolfgang; Beck, Erich;

Jaworek, Thomas; Koeniger, Rainer

PATENT ASSIGNEE(S):

SOURCE:

Basf A.-G., Germany Ger. Offen., 6 pp.

CODEN: GWXXBX

DOCUMENT TYPE:

LANGUAGE:

Patent German

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO. DATE
DE 19929258	A1	20001228	DE 1999-19929258 19990625
US 6458991	В1	20021001	US 2000-592736 20000613
JP 2001026569	A2	20010130	JP 2000-187845 20000622
EP 1063225	A2	20001227	EP 2000-113365 20000623
EP 1063225	Α3	20020717	

R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO

PRIORITY APPLN. INFO.:

DE 1999-19929258 A 19990625

Compds. with alc. OH groups are esterified, in the presence of Cu(II) salts or their mixts. with Cu(I) salts as polymerization inhibitors, with acrylic or methacrylic acid in the presence of a hydrocarbon (b.p. 60-140°) for azeotropic removal of water at 90-150°. After esterification the Cu salt is precipitated as CuS and the ester is separated The resulting light-colored esters, obtained in high yields, are particularly suitable for colored pigmented coatings on substrates such as wood, paper, mineral building materials, plastic or metal.

L16 ANSWER 21 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN

1999:34537 CAPLUS ACCESSION NUMBER:

130:95963 DOCUMENT NUMBER:

Zirconium catalyzed transesterification TITLE: process for preparing synthetic wax

monomers

Schlaefer, Francis William; Gross, Andrew William INVENTOR (S):

Rohm and Haas Company, USA PATENT ASSIGNEE(S):

U.S., 5 pp. SOURCE: CODEN: USXXAM

Patent DOCUMENT TYPE: English LANGUAGE:

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE PATENT NO. KIND DATE \_\_\_\_\_ 19990105 US 1997-950443 19971015 US 1997-950443 19971015 US 5856611 A PRIORITY APPLN. INFO.:

A zirconium catalyzed transesterification process for

preparing synthetic wax monomers comprises reacting a synthetic wax

alc. with a (meth) acrylic acid ester

in the presence of a zirconium catalyst and an inhibitor.

monomers are useful in a wide range of polymers.

THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 4 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L16 ANSWER 22 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 20

ACCESSION NUMBER: 1998:421476 CAPLUS

DOCUMENT NUMBER: 129:82071

Manufacture of tetrahydrobenzyl (meth) acrylate by TITLE:

using polymerization inhibitors and oxygen

gas for reaction efficiency

Fujiwara, Keisuke INVENTOR(S):

Daicel Chemical Industries, Ltd., Japan PATENT ASSIGNEE(S):

Jpn. Kokai Tokkyo Koho, 7 pp. SOURCE:

CODEN: JKXXAF

Patent DOCUMENT TYPE: Japanese LANGUAGE:

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE JP 10175919 A2 19980630 JP 1996-353733 19961218 RITY APPLN. INFO.: JP 1996-353733 19961218 PRIORITY APPLN. INFO.:

OTHER SOURCE(S): MARPAT 129:82071

Tetrahydrobenzyl alc. (I) is esterified with ( meth) acrylic acid in the presence of 02-containing

gases and polymerization inhibitors of (1) (A) quinones, hindered phenols, nitrosoamines, and/or phenylenediamines or (B) phenothiazines, RR'NOH (R, R' = H, alkyl, aryl), Cu(S2CNR12)2 (R1 = alkyl, aryl), and/or Fe(S2CNR12)3 or (2) (A') hydroquinone, hydroquinone monomethyl ether (II),

benzoquinone, 3,5-di-tert-butyl-4-hydroxytoluene (III),

N-nitrosodiphenylamine, and/or N,N'-diphenylphenylenediamine or (B')

phenothiazine, Cu(S2CNMe2)2 (IV), Cu(S2CNEt2)2, Cu(S2CNPr2)2, Fe(S2CNMe2)3, and/or Et2NOH, and the resulting crude solution is distilled with 02-containing gases and the above polymerization inhibitors for purification Alternatively, the esterification is carried out by using the

gases and the inhibitors of A', and the resulting solution is distilled with the gases and the inhibitors of B. Thus, a solution of I was bubbled with air and reacted with methacrylic acid in the presence of a catalyst and polymerization **inhibitors** of II and III. Then, the product solution after catalyst removal was refluxed with IV to give tetrahydrobenzyl methacrylate with yield 84%.

L16 ANSWER 23 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 21

ACCESSION NUMBER: 1998:421475 CAPLUS

DOCUMENT NUMBER: 129:68146

TITLE: Production of tetrahydrobenzyl (meta)acrylate in

simple condition for improved reaction efficiency

INVENTOR(S): Fujiwara, Keisuke

PATENT ASSIGNEE(S): Daicel Chemical Industries, Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 10175918 A2 19980630 JP 1996-353732 19961218

PRIORITY APPLN. INFO.: JP 1996-353732 19961218

OTHER SOURCE(S): MARPAT 129:68146

AB Tetrahydrobenzyl alc. (I) is esterified with (

meth)acrylic acid in the presence of polymerization inhibitors and catalysts in dehydration solvents of C≤10 hydrocarbons to carry out dehydration by azeotropic distillation Catalysts are removed from the resulting crude solution to attain catalyst concentration

semoved from the resulting clade solution to detail energy  $\leq 10\%$ , and then the crude solution is distilled for purification. Thus, I in n-hexane was reacted with methacrylic acid in the presence of

methanesulfonic acid (catalyst) and polymerization inhibitors. The product solution was heated, stirred with water, and separated to organic and aqueous

layers, so that 98.3% (based on initial catalyst content) of the catalyst was removed. Then, the resulting solution was distilled with another polymerization

inhibitor to give tetrahydrobenzyl methacrylate with yield 84%.

L16 ANSWER 24 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 22

ACCESSION NUMBER: 1998:126650 CAPLUS

DOCUMENT NUMBER: 128:193033

TITLE: Preparation of (meth)acrylate esters

INVENTOR(S): Okada, Shinji; Muraue, Takao; Mishina, Hiroya PATENT ASSIGNEE(S): Toa Gosei Chemical Industry Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 10053559 A2 19980224 JP 1996-227883 19960809
PRIORITY APPLN. INFO.: JP 1996-227883 19960809

AB Title esters are **prepared** by reacting (meth)

acrylic acids and alcs. in the presence of

catalysts with removing H2O by distillation, in which aqueous basic solns. are added

to the distillate for adjusting pH at ≥6.5. Corrosion on the reactors caused by the distillates or polymerization of the (meth) acrylic acid in the distillates are avoided in the process. Thus, acrylic acid, BuOH, H2\$O4, hexane, and polymerization inhibitor were continuously charged at the rates of 100, 115, 1.4, 56, and 0.3 part/h, resp., at 120° into a reactor equipped with (a) a distillation column connected with a condenser and with (b) a separator from which the organic layer is recycled to a distillation column. Crude ester and

were continuously discharged from the reactor bottom and from the separator, resp., with continuously charging 25%-NaOH into the vapor from the distillation column in order to maintain pH at 7-8 at the separator. The reaction could be continued for 100 days with no polymer formation in the separator and corrosion of the SUS-304 condenser.

L16 ANSWER 25 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 23

ACCESSION NUMBER:

1998:165487 CAPLUS

DOCUMENT NUMBER:

128:193025

TITLE:

Manufacture of halogenated (meth) acrylic esters and poly (meth)

acrylates

INVENTOR(S):

Hofstraat, Johannes Willem; Wakselman, Claude; Lequesne, Christelle; Wiersum, Ulfert Elle;

Blazejewski, Jean Claude

PATENT ASSIGNEE(S): SOURCE:

Akzo Nobel N.V., Neth. Eur. Pat. Appl., 21 pp.

CODEN: EPXXDW

DOCUMENT TYPE:

LANGUAGE:

Patent English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

OTHER SOURCE(S):

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 824096	A2	19980218	EP 1997-202331	19970729
EP 824096	A3	19980408		an wa nm
R: AT, BE,	CH, DE	, DK, ES, FR	, GB, GR, IT, LI, LU	, NL, SE, MC, PT,
IE, FI				
CA 2212733	AA	19980213	CA 1997-2212733	19970811
JP 10095751	A2	19980414	JP 1997-230482	19970813
PRIORITY APPLN. INFO	. :		EP 1996-202276	19960813

AB The present invention is directed to the **preparation** of (meth) acrylates of halogenated **alcs**. by direct

esterification of the alcs. with (meth) acryloyl chloride

MARPAT 128:193025

wherein at least one 2,6-substituted pyridine derivative is used as a polymerization

inhibitor at a temperature <50°. Suitable 2,6-substituted
pyridine derivs. are 2,6-lutidine, 2,4,6-collidine or 2,6-di-tert-butyl-4methylpyridine. With the halogenated (meth)acrylate esters prepd
. according to the invention, very pure homo- and copolymers can be
prepared, resulting in polymers having a low optical loss of
≤0.1 dB/cm at 1300 nm and <0.4 dB/cm at 1550 nm. The polymers are
useful in making a waveguide.</pre>

L16 ANSWER 26 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 24

ACCESSION NUMBER: 1998:808209 CAPLUS

DOCUMENT NUMBER: 130:223626

TITLE: Study on catalytic properties and polymerization

retardation of heteropoly acid in the

preparation of acrylate

AUTHOR(S):

Ma, Liqun; Yang, Yulin

CORPORATE SOURCE:

Department of Chemical Engineering, Industry

Institute, Qiqihar University, Qiqihar, 161006, Peop.

Rep. China

SOURCE:

Huagong Shikan (1998), 12(10), 16-18

CODEN: HUSHFT; ISSN: 1002-154X

Huagong Shikan Zazhishe PUBLISHER:

DOCUMENT TYPE:

Journal

Chinese LANGUAGE:

Ethylene glycol diacrylate, tetrahydrofurfuryl methacrylate and octadecyl

methacrylate were prepared by esterification in the

presence of heteropoly acid catalyst. Suitable heteropoly acid-

alc. ratio was 1.0-1.5% and suitable (meth) acrylic acid-alc. was 2.2.apprx.2.4:1. The

heteropoly acid showed good polymerization inhibition effect for Me methacrylate.

L16 ANSWER 27 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 25

ACCESSION NUMBER: DOCUMENT NUMBER:

1997:754319 CAPLUS

TITLE:

128:23268

Preparation of (meth)

acrylic acid esters with

polymerization prevention and apparatus therefor Okata, Shinji; Muraue, Takao; Mishina, Hiroya Toa Gosei Chemical Industry Co., Ltd., Japan

SOURCE:

Jpn. Kokai Tokkyo Koho, 6 pp. CODEN: JKXXAF

DOCUMENT TYPE:

INVENTOR(S):

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT ASSIGNEE(S):

TR 00203037 PATENT NO. JP 1996-140966 19960510 JP 09301921 A2 19971125 JP 1996-140966 19960510 PRIORITY APPLN. INFO.:

(Meth) acrylic acids are esterified

with alcs. in the presence of catalysts to give crude products, which are added with polymerization inhibitors and Michael addition products of (meth) acrylic acid esters, and

remove low-b.p. impurities by distillation The Michael addition products are added

so that their concentration in resulting residue liqs. being 3-30%. heating 100 parts acrylic acid with 152 parts 2-ethylhexyl alc. at 120° in the presence of H2SO4 and phenothiazine (I) in PhMe gave a crude ester liquid, which was purified by distillation at 135-145° in the presence of I, 2-ethylhexyl  $\beta$ -2-ethylhexyloxypropionate, and 2-ethylhexyl  $\beta$ -acryloxypropionate to give 2-ethylhexyl acrylate in high yield.

L16 ANSWER 28 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 26

1995:737585 CAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER:

123:143297

TITLE:

Process for the transesterification

of (meth) acrylic acid

esters.

INVENTOR(S):

Knebel, Joachim; Pfirmann, Martina; Ohl, Thomas

Rohm GmbH, Germany PATENT ASSIGNEE(S):

Eur. Pat. Appl., 10 pp. SOURCE:

CODEN: EPXXDW

DOCUMENT TYPE:

Patent German

LANGUAGE:

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE EP 663386 A1 19950719
FD 663386 B1 19990414 EP 1995-100110 19950105 19950719 R: AT, BE, CH, DE, DK, ES, FR, GB, IT, LI, NL, SE DE 4401132 A1 19950720 DE 1994-4401132 19940117 AT 178880 E 19990415 AT 1995-100110 19950105 CA 2140304 AA 19950718 CA 1995-2140304 19950116 RITY APPLN. INFO.: DE 1994-4401132 A 19940117 PRIORITY APPLN. INFO.: OTHER SOURCE(S): CASREACT 123:143297; MARPAT 123:143297 Acrylate and methacrylate esters are **prepared** in higher yield and purity by **transesterification** of lower alkyl (meth) acrylates CH2:C(R1)CO2R2 [R1 = H, Me; R2 = C1-6 alkyl] with mono- or polyhydric alcs. in the presence of mixed catalysts consisting of 5-95% diorganyltin oxide R3R4Sn:O [R3, R4 = C1-12 aliphatic, aromatic, or araliph. group] and 95-5% organotin halide R3R4SnX2 or R3SnX3 [X = Cl, Br, iodo, cyano, isocyanato, isothiocyanato]. For example, transesterification of triethylene glycol with excess Me methacrylate in the presence of equal amts. of Bu2Sn:O and Bu2SnCl2, plus 4-HOC6H4OMe (polymerization inhibitor), under reflux with distillation of MeOH and Me methacrylate, gave triethylene glycol dimethacrylate in 87% yield, and containing only 0.8% monomethacrylate byproduct and only 1.3% Me methacrylate. In contrast, a run with only Bu2SnCl2 catalyst gave no reaction. Similarly prepared in good yields were ethylene glycol dimethacrylate, pentaerythritol tetramethacrylate and tetraacrylate, and tetrahydrofurfuryl methacrylate.

L16 ANSWER 29 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 27

ACCESSION NUMBER: 1995:207908 CAPLUS

DOCUMENT NUMBER:

122:82299

TITLE:

Manufacture of (meth)acrylic

acid esters

INVENTOR(S):

Ri, Shotaku; Takahashi, Katsuji; Okuda, Ryuji

PATENT ASSIGNEE(S):

Dainippon Ink & Chemicals, Japan Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

SOURCE:

Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE JP 06234700 A2 19940823 JP 1993-24124 19930212 RITY APPLN. INFO.: JP 1993-24124 19930212 PRIORITY APPLN. INFO.:

The compds. are manufactured by esterification of (meth) acrylic acid with alcs. in the presence of polymerization inhibitors and esterification catalysts, ≥1 of which being water-soluble; and washing the reaction mixture with H2O to remove the water-soluble polymerization inhibitors and/or catalysts. Thus, heating acrylic acid with trimethylolpropane in the presence of 2-methyl-1-phenol-4-sulfonic acid, Cu sulfate, and p-MeC6H4SO3H in PhMe at 65° and 130 torr for 10 h, and washing the reaction mixture with H2O gave 99.5% acrylic esters with 100% removal of the polymerization inhibitors and the catalyst.

L16 ANSWER 30 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 28

ACCESSION NUMBER:

1995:207907 CAPLUS

DOCUMENT NUMBER:

122:82298

TITLE:

Manufacture of (meth) acrylic

acid esters

INVENTOR(S):

Ri, Shotaku; Takahashi, Katsuji; Okuda, Ryuji

PATENT ASSIGNEE(S):

Dainippon Ink & Chemicals, Japan Jpn. Kokai Tokkyo Koho, 8 pp.

SOURCE:

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 06234699	A2	19940823	JP 1993-24125	19930212
ORITY APPLN. INFO.	:	JР	1993-24125	19930212
		/.	\	_£ /

PRIO The compds. are manufactured by (trans)esterification of ( AB

meth) acrylic acid esters with alcs.

in the presence of water-soluble polymerization inhibitors and water-soluble catalysts, washing with H2O, and recycling of the polymerization inhibitors and the catalysts. Thus, acrylic acid was heated with trimethylolpropane in the presence of 2-methyl-1-phenol-4-sulfonic acid, Cu sulfate, and p-MeC6H4SO3H in PhMe at 65° and 130 torr for 10 h, and washed with H2O twice to give 99.5% acrylic acid ester. The polymerization inhibitors and the catalyst in the 1st washing water were recycled and used for the reaction giving 99.5% acrylate.

L16 ANSWER 31 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 29

ACCESSION NUMBER:

1994:410191 CAPLUS

DOCUMENT NUMBER:

121:10191

TITLE:

Manufacture of (meth)acrylates

INVENTOR(S):

Takahashi, Katsuji; Tani, Juichiro; Ri, Shotaku;

Okuda, Tatsushi

PATENT ASSIGNEE(S): SOURCE:

Dainippon Ink & Chemicals, Japan Jpn. Kokai Tokkyo Koho, 13 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 06016594	A2	19940125	JP 1993-55628	19930316
JP 2914075	B2	19990628		
PRIORITY APPLN. INFO.	:		JP 1992-111414	19920430

(Meth) acrylates are synthesized by esterification of ( meth) acrylic acids with alcs. or by

transesterification of (meth) acrylates with alcs. in the presence of a compound which acts as both an esterification catalyst and a polymerization inhibitor. Compds. which can be used for this purpose are aromatic sulfonic acids with phenolic hydroxy, amino, nitro or nitroso groups and/or partially transition metal ion-exchanged strongly acidic cation-exchange resins. For example, 201 g trimethylolpropane and 422 g acrylic acid reacted in the presence of 12.5 g hydroquinone-2sulfonic acid in cyclohexane-toluene with 98.6% alc. conversion

and 95.6% ester yield; no polymeric species was detected.

L16 ANSWER 32 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 30

ACCESSION NUMBER: 1994:410190 CAPLUS

121:10190 DOCUMENT NUMBER:

Preparation of (meth) TITLE: acrylic acid esters

Tani, Juichiro; Okuda, Ryuji; Takahashi, Katsuji; Ri, INVENTOR(S):

Shotaku

Dainippon Ink & Chemicals, Japan PATENT ASSIGNEE(S): Jpn. Kokai Tokkyo Koho, 19 pp. SOURCE:

CODEN: JKXXAF

DOCUMENT TYPE: Patent Japanese LANGUAGE:

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 06009496 A2 19940118 JP 1993-55627 19930316
PARTENT NO. JP 1992-111415 19920430 PRIORITY APPLN. INFO.:

The title compds. are easily **prepared** by reacting (meth)

acrylic acid with alcs. in the presence of water-soluble esterification catalysts and water-soluble polymerization inhibitors, then washing the products with water to remove the catalysts and inhibitors. The uses of these catalysts and polymerization inhibitors simplifies the workup process. Thus, heating trimethylolpropane 201, acrylic acid 422, Na hydroquinonesulfonate 3.1, and p-toluenesulfonic acid 12.5 g in 13 g PhMe and 112 g cyclohexane at 100° for 6 h and working up gave a corresponding ester at 95.8% yield.

L16 ANSWER 33 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 31

ACCESSION NUMBER: 1994:165213 CAPLUS

120:165213 DOCUMENT NUMBER:

Preparation of acrylic ester or methacrylic TITLE:

ester

Haga, Masami INVENTOR(S):

Idemitsu Petrochemical Co, Japan PATENT ASSIGNEE(S): Jpn. Kokai Tokkyo Koho, 7 pp. SOURCE:

CODEN: JKXXAF

Patent DOCUMENT TYPE: Japanese LANGUAGE:

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 05320095 A2 19931203 JP 1992-128363 19920521
RITY APPLN. INFO.: JP 1992-128363 19920521 PRIORITY APPLN. INFO.: The title esters without discoloration are prepared by treating ( meth) acrylic acid with alcs. in the

presence of acid catalysts and aromatic compds. having  $\geq 1$  alkyl and ≥2 OH (per 1 benzene ring) as polymerization inhibitors. Thus, 1.20 mol acrylic acid was esterified with 1.00 mol 2-ethylhexyl

alc. (I) in cyclohexane in the presence of 50 mg

2-tert-butylhydroquinone and 0.50 g concentrated H2SO4 while bubbling with N having O content 6 volume% at 84-98° for 6 h to give an ester with ASTM hue L value 0.5 without polymer generation at 98.9 mol% I conversion.

L16 ANSWER 34 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 32

1993:473881 CAPLUS ACCESSION NUMBER:

119:73881 DOCUMENT NUMBER:

Curable (meth)acrylate compositions with low viscosity TITLE:

and manufacture thereof

Yaqi, Hirobumi; Sugimura, Toshiro INVENTOR(S): Nippon Shokubai Co., Ltd., Japan PATENT ASSIGNEE(S):

Jpn. Kokai Tokkyo Koho, 7 pp. SOURCE:

CODEN: JKXXAF

Patent DOCUMENT TYPE: Japanese LANGUAGE:

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

APPLICATION NO. DATE KIND DATE PATENT NO. \_\_\_\_\_ \_\_\_\_ \_\_\_\_\_ JP 04353503 A2 JP 3107851 B2 19921208 JP 1991-129715 19910531 20001113

JP 2000103815 A2 :ITY APPLN. TNPO JP 1999-205291 19910531 20000411 JP 1991-129715 A3 19910531 PRIORITY APPLN. INFO.:

The title composition usable with high filler content and in photocurable epoxy resist inks contain inorg. materials and (meth)acrylates prepared from (meth) acrylic acid and/or

(meth) acrylates and alcs. in the presence of a Cu compound A composition from 80 g trimethylolpropane triacrylate prepared from trimethylolpropane and acrylic acid in the presence of p-toluenesulfonic acid and Cu20 and 20 g talc had viscosity (25°) 900 cP, compared with 1700 cP for a control using hydroquinone polymerization inhibitor in place of Cu20.

L16 ANSWER 35 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 33

ACCESSION NUMBER: 1994:580428 CAPLUS

121:180428 DOCUMENT NUMBER:

Process for preparing TITLE:

fluorohaloalkyl (meth)acrylates Chvatal, Zdenek; Dedek, Vaclav; Mazac, Jiri; Marousek, INVENTOR(S):

Vladimir; Bednar, Bohumil

Vysoka Skola Chemicko-Technologicka, Czech. PATENT ASSIGNEE(S):

SOURCE:

Czech., 4 pp. CODEN: CZXXA9

Patent

DOCUMENT TYPE: Czech LANGUAGE:

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

APPLICATION NO. DATE KIND DATE PATENT NO. ---------\_\_\_\_\_ CS 1989-2377 CS 276012 B6 19920318 19890417 CS 1989-2377 19890417 PRIORITY APPLN. INFO.: MARPAT 121:180428 OTHER SOURCE(S):

CH2:CRCO2(CH2)nX (I; R = H, Me; X = BrCF2CF2, ClCF2CF2, BrCF2CFC1, ClCF2CFC1, 1-chlorotrifluoroethyl, BrCF2CFCF3, F3CCFBrCF2, F3CCBrCF3; n =

1-4) were prepared by acid-catalyzed esterification of

the appropriate alc. with (meth) acrylic

acid in the presence of polymerization inhibitor or by esterification of the acid chloride in an alkaline medium in the presence of a phase-transfer catalyst. For example, 6 g CH2:CMeCOCl was added dropwise at 0° to a stirred mixture of 10 g BrCF2CF2CH2CH2OH, 3 g KOH, 0.32 g Bu4NBr, 50 mL CH2Cl2, and 50 mL H2O and the whole stirred for 2.5 h to give 8.5 g of 97%-pure (GLC) I (R = Me, X = BrCF2CF2, n = 2). L16 ANSWER 36 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 34

ACCESSION NUMBER:

1991:409589 CAPLUS

DOCUMENT NUMBER:

115:9589

TITLE:

Manufacture of (meth) acrylate esters

INVENTOR(S):

Haga, Masami

PATENT ASSIGNEE(S):

Idemitsu Petrochemical Co., Ltd., Japan

SOURCE:

Jpn. Kokai Tokkyo Koho, 9 pp. CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE PATENT NO. KIND DATE \_\_\_\_\_ \_\_\_\_\_ JP 1989-166565 19890630 JP 03034956 A2 JP 07064791 B4 19910214

B4 19950712

JP 1989-166565 19890630

PRIORITY APPLN. INFO.:

In preparation of (meth) acrylate esters by reaction of (meth

) acrylic acid with alcs. in solvents in the presence of acid catalysts, byproduct formation is reduced by (A) using ( meth) acrylic acids containing polymerization

inhibitors, (B) carrying out the reaction under inert gas containing 0.1-10 volume% O, and optionally (C) treating the reaction mixts. with aqueous alkali at 40-95° and washing the separated oily phase with H2O at 40-95°. Thus, stirring a mixture of 186.3 g n-dodecyl alc

., 90.4 g methacrylic acid, 0.5 g concentrate H2SO4, 50 mg methoxyhydroquinone, and 80 mL PhMe under N containing 6.0 volume% O at 125-151°, mixing the reaction mixture with aqueous NaOH at 60° for 5 min, separating the oily phase, washing with H2O at 60° for 5 min, and distillation of the residue gave 98.7 mol% n-dodecyl methacrylate of 98.9% purity.

L16 ANSWER 37 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 35

ACCESSION NUMBER:

1991:7402 CAPLUS

DOCUMENT NUMBER:

114:7402

TITLE:

Manufacture of (meth)acrylate esters of polyhydric

alcohols having reduced color

INVENTOR(S):

Ritter, Wolfgang; Sitz, Hans Dieter; Speitkamp, Ludwig

Henkel K.-G.a.A., Germany

SOURCE:

Ger. Offen., 6 pp.

CODEN: GWXXBX

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT ASSIGNEE(S):

PATENT NO.	KIND DATE	APPLICATION NO.	DATE
DE 3843938	A1 19900628	DE 1988-3843938	19881224
EP 376091	A1 19900704	EP 1989-123220	19891215
R: GR			
WO 9007487	A1 19900712	WO 1989-EP1550	19891215
W: JP, US			
RW: AT, BE,	CH, DE, ES, FR,	GB, IT, LU, NL, SE	
EP 449912	A1 19911009	EP 1990-900819	19891215
EP 449912	B1 19940302		
R: BE, CH,	DE, ES, FR, GB,	IT, LI, NL	
JP 04502468	T2 19920507	JP 1990-501716	19891215
CA 2006434	AA 19900624	CA 1989-2006434	19891221

US 5198574 A 19930330 US 1991-720445 19910624
PRIORITY APPLN. INFO.: DE 1988-3843938 19881224
WO 1989-EP1550 19891215

The title monomers are prepared by the esterification of polyhydric alcs. with (meth) acrylic acids in the presence of nonsubstituted phenol polymerization inhibitors, transesterification catalysts, and activated charcoal color reducing agent. Thus, acrylic acid 928.8, propoxylated neopentyl glycol (OH value 416 mg KOH/g) 1560.4, p-toluene sulfonic acid 87.1, activated charcoal 124.5 and hydroquinone (1100 ppm based on product mixture) 2.5 g were mixed together under a air/N mixture (5 volume% O; 20 L/h) at 135-143° for 5 h, then the activated carbon was removed by filtration, producing a crude product having acid value 34 mg KOH/g, OH value 10 mg KOH/g, Gardner color value <1, viscosity 92 mPa-s, which was mixed with 4 L of aqueous 16% NaCl and 4% NaHO3 solution in the presence of 200 ppm hydroquinone monomethyl ether and dried at 80°/40 mBar for 3 h, and filtered, producing a product having acid value <1 mg KOH/g, OH value <15 mg KOH/g and Gardner color value 3, vs. <1, <15, and 8-9 for a control product prepared without the addition of activated charcoal.

L16 ANSWER 38 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 36

ACCESSION NUMBER: 1991:7401 CAPLUS

DOCUMENT NUMBER: 114:7401

TITLE: Solventless manufacture of (meth) acrylate esters of

polyhydric alcohols

INVENTOR(S): Ritter, Wolfgang; Sitz, Hans Dieter; Speitkamp, Ludwig

PATENT ASSIGNEE(S): Henkel K.-G.a.A., Germany

SOURCE: Ger. Of

Ger. Offen., 8 pp. CODEN: GWXXBX

DOCUMENT TYPE:

TYPE: Patent German

LANGUAGE: Ger FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PF	TENT 1	NO.		KIN	1D	DATE			AI	PLI	CAT	ION	NC.	).	DATE	
DF	3843	 930	<b>-</b>	 A1	 L	1990	0628		DI	 3 19	 988-	384	393	0	19881224	<u>.</u>
	3760	90		A1	L	1990	0704		EI	2 19	989-	123	219	)	19891215	5
WC	R: 9007			A1	L	1990	0712		WC	) 19	989-	EP1	548	3	19891215	5
		JP,						~-				~				
	RW:	ΑT,	BE,	CH,	DE,	, ES,	FR,	GB,	II,	ьU,	, NL	ı, S	) E			_
EI	4499	13		A1	L	1991	1009		E	2 19	990-	900	821	-	19891215	)
EI	4499	13		B1	L	1994	0309									
	R:	BE.	CH,	DE,	ES.	FR,	GB,	IT,	LI,	NL						
JI	0450			T2			0507				990-	501	714	Ł	19891215	ō
JI	3036	832		В2	2	2000	0424									
	2050			$T^3$	3	1994	0516		E	3 19	990-	900	821	L	19891215	5
	2006			A.	A	1990	0624		CZ	A 19	989-	200	643	31	19891223	l
	5648			A		1997	0715		US	3 19	991-	679	075	5	19910823	3
PRIORI		-	TNFO	. :					DE 19	988-	-384	393	0	Α	19881224	4
FREORE.			11.1					1	WO 19	989-	-EP1	548	3	W	19891215	5
AD TO	. +i+	1 a m.	onome	are	ar.	nre	nare	a hv	the	eat	teri	fic	ati	on		

The title monomers, are prepared by the esterification of (meth) acrylic acid in the presence of acidic esterification catalysts and in the presence of α-tocopherol as a polymerization inhibitor. Reaction water is removed from the reaction mixture under elevated temps., producing a reaction mixture without a solvent which would otherwise need to be removed or could form an azeotropic-forming agent. Thus, acrylic acid 324.0, ethoxylated trimethylolpropane (OH value 680 mg KOH/g) 368.2,

p-toluenesulfonic acid 24.2, and D,L- $\alpha$ -tocopherol 1.38 g were heated together at 145° while an air/N mixture (5 volume% O, 40 L/h) was passed through the reaction mixture, the crude product was washed with 700 mL 10% aqueous Na2CO3 solution and dried in vacuum at 80°/40 mbar for 3 h, producing ethoxylated trimethylolpropane acrylates having acid value <1 mg KOH/g, OH value 43 mg KOH/g, Gardener color value 3-4, and H2O content 0.31%.

L16 ANSWER 39 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 37

ACCESSION NUMBER:

1990:632240 CAPLUS

DOCUMENT NUMBER:

113:232240

TITLE:

Process for the manufacture of

(meth) acrylate esters of polyhydric alcohols

Ritter, Wolfgang; Sitz, Hans Dieter; Speitkamp, Ludwig

INVENTOR(S): Ritter, Wolfgang; Sitz, Haraman Assignee(S): Henkel K.-G.a.A., Germany

SOURCE:

Ger. Offen., 5 pp. CODEN: GWXXBX

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PAT	ENT NO.		KIND	DATE	APPLICATION NO.	DATE
DE	3843854		A1	19900628	DE 1988-3843854	19881224
EP	376088		A1	19900704	EP 1989-123217	19891215
	R: GF	2				
WO	9007486	;	A1	19900712	WO 1989-EP1549	19891215
	W: JI	, US				
	RW: A	, BE,	CH, DE	, ES, FR,	GB, IT, LU, NL, SE	
EP	449918		A1	19911009	EP 1990-900856	19891215
ΕP	449918			19941005		
	R: BE	CH,	DE, ES	, FR, GB,	IT, LI, NL	
JP	0450246	7	T2	19920507	JP 1990-501715	19891215
ES	2061009	)	<b>T</b> 3	19941201	ES 1990-900856	19891215
JP	2935742	:	B2	19990816	JP 1989-501715	19891215
CA	2006433	}	AA	19900624	CA 1989-2006433	19891221
US	5350877	,	Α	19940927	US 1991-720443	19910624
PRIORITY	APPLN.	INFO	.:		DE 1988-3843854 A	19881224
					WO 1989-EP1549 W	19891215

AB The title compds. are prepared by the reaction of (meth)

acrylic acid in the presence of

transesterification catalysts and substituted phenolic polymerization inhibitors with polyhydric alcs. so as to produce a reaction mixture which is substantially free of solvents and/or azeotropic carriers, and the produced reaction water is removed from the reaction mixture in the gas phase. Thus, acrylic acid 1559.5, ethoxylated trimethylolpropane (I, OH value 680) 1521.0, p-toluenesulfonic acid 107.8, and 2,5-di-tert-butylhydroquinone 4.96 g were contacted with 40 L/h of air to remove water, esterified for 6 h, heated at 105°/400 mbar for 6 h, neutralized with 103 g Ca(OH)2, stirred at 80°/50 mbar, and filtered to produce I acrylate having acid value <1, OH value 24, and Gardner color >1.

L16 ANSWER 40 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 38

ACCESSION NUMBER: 1991:24763 CAPLUS

114:24763

TITLE:

DOCUMENT NUMBER:

Process for the manufacture of

(meth)acrylate esters of polyhydric alcohols

INVENTOR(S): Ritter, Wolfgang; Sitz, Hans Dieter; Speitkamp, Ludwig

PATENT ASSIGNEE(S):

Henkel K.-G.a.A., Germany

SOURCE:

Ger. Offen., 9 pp. CODEN: GWXXBX

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PAT	TENT NO.		KIND	DATE		APPLICATION NO. DATE
	3843843					DE 1988-3843843 19881224
	376089			19900704		EP 1989-123218 19891215
	R: GR					
EP	377156		A1	19900711		EP 1989-123221 19891215
	R: GR					WO 1989-EP1546 19891215
WO			A1	19900712		WO 1989-EP1546 19891215
	W: JP,		au DB	TO ID	an .	rm tit Nit CE
	-		CH, DE,	10000712	GB, .	IT, LU, NL, SE WO 1989-EP1547 19891215
WO	9007484		ΑŢ	19900/12		WO 1989-EP1547 19891215
	W: JP,	US	CII DE	EC ED	an .	TO THE MI. CD
		BE,	CH, DE,	10011000	GB,	IT, LU, NL, SE EP 1990-900162 19891215
EP	449865 449865		A1	19911009		EP 1990-900102 19091213
EP						TT NT
			DE, ES,	FR, GB,	11,	EP 1990-900865 19891215
	449919					EP 1990-900003 19091213
EP	449919	ar.	DD DG	19940504	TOT .	T.T. MT.
			T2	FR, GB,	тт, .	JP 1990-500801 19891215
JP	04502455		T2	19920507		JP 1990-501465 19891215
JP	04502459		172	19920307		ES 1990-900865 19891215
ES	20531/1		T3	19940716		JP 1989-501465 19891215
JP	2758717		B2	10000303		
	2863309		B∠ 7.7	19990303		<b>01 200 3</b>
	2006430		AA	19900624 19900624		CA 1989-2006430 19891221 CA 1989-2006432 19891221
	2006432 618875		AA B2	19900024		AU 1990-49716 19900212
AU	9049716		D∠ 7\1	19910829		A0 1550 45710 15500212
			B2			AU 1990-49718 19900212
	9049718			19910905		110 1330 13710 23301===
			B2	19920514		AU 1990-49722 19900212
	9049777		Δ1	19910829		
אט	9049722 626986		('L	19920813		AU 1990-49721 19900212
	9049721		A1	19910829		
	5159106		Α			US 1991-679073 19910820
	5210281		A			US 1991-720444 19910823
	Y APPLN.				D	E 1988-3843843 19881224
INTORIT			-		D	E 1989-3939163 19891127
					W	E 1989-3939163 19891127 O 1989-EP1546 19891215
					W	O 1989-EP1547 19891215

AB In the title process, (meth) acrylic acids are esterified with polyhydric alcs. in the presence of acidic esterification catalysts and liquid droplets of polymerization inhibitor. The crude product may then be neutralized with oxides and/or hydroxides of alkaline earth metals and/or Al, and treated with decoloring agents. Thus, acrylic acid 14.53, ethoxylated trimethylolpropane (OH value 665 mm-KOH/g) 14.18, p-toluenesulfonic acid 1.01, and 2,5-di-tert-butylhydroquinone 0.047 kg were contacted with 100 L/h of air, the product stream heated to 105° and contacted with an air stream (60 L/h), water removed under the following vacuum profile: 2 h at 105°/400 mbar, 1 h at 105°/300 mbar, 0.5 h at 105°/200 mbar, 1 h at 105°/100 mbar, and 0.5 h at

105°/23 mbar. The intermediate was mixed with 0.53 kg Ca(OH)2, stirred at 80°/50 mbars, and filtered, producing polyethylene glycol trimethylolpropane ether acrylate having acid value <1 mg-KOH/g, OH value 14 mg-KOH/g, and Gardner color value <1.

L16 ANSWER 41 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 39

ACCESSION NUMBER:

1990:119592 CAPLUS

DOCUMENT NUMBER:

112:119592

TITLE:

Manufacture of polyfunctional (meth) acrylate esters

with little discoloration

INVENTOR (S):

Honma, Akihiro

PATENT ASSIGNEE(S):

Mitsubishi Gas Chemical Co., Inc., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 4 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 01258643	A2	19891016	JP 1988-83104	19880406
JP 2586092	B2	19970226		

PRIORITY APPLN. INFO.:

JP 1988-83104 19880406

AB Light-colored transparent title esters are manufactured by **esterifying** (meth)acrylic acid with polyhydric

alcs. in the presence of an acid catalyst and pyrogallol in contact with O2-containing gas, then removing the catalyst by extraction with

neutralizing excess (meth) acrylic acid with

an aqueous alkaline solution, and washing. Thus, a mixture of methacrylic

trimethylolpropane 148, p-MeC6H4SO3H 34, pyrogallol 0.6, and cyclohexane 130 g was refluxed while bubbling air through the solution, and distilling off 61.5 g H2O. Then 242 g cyclohexane and 100 g H2O were added, the aqueous layer containing p-MeC6H4SO3H was separated, the organic layer was neutralized

contacting with 372 g 15% aqueous NaOH, then washing with H2O, adding 0.01 g toluhydroquinone, and distilling in vacuo to give 350 g trimethylolpropane trimethacrylate with APHA color 20, vs. 60 using hydroquinone instead of pyrogallol.

L16 ANSWER 42 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 40

ACCESSION NUMBER:

1989:407990 CAPLUS

DOCUMENT NUMBER:

111:7990

TITLE:

H20,

by

Manufacture of esters of unsaturated acids with cation

exchangers as catalysts

INVENTOR(S):

Nakajima, Sumio; Sogabe, Hideki; Yoshida, Hiroshi;

Okubo, Atsushi

PATENT ASSIGNEE(S):

Nippon Shokubai Kagaku Kogyo Co., Ltd., Japan

Jpn. Kokai Tokkyo Koho, 6 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

SOURCE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 01009956	A2	19890113	JP 1987-162251	19870701

JP 1987-162251 PRIORITY APPLN. INFO.: Esters are prepared in high yield with min. byproduct formation by esterification of (meth) acrylic acid with C4-12 aliphatic alcs. in the presence of strongly acid cation exchange resins as catalysts with stirring at 0.005-2 kW/m3 in boiling alc. solution containing water, the water being removed with alc . and the unreacted acids being recovered in the organic phase. A mixture of acrylic acid (I) 49, 2-ethylhexanol (II) 36.6, 2-ethylhexyl acrylate 14.2, and water 0.1% was added to a reactor containing 14 L ion exchange resin (Diaion PK-208) to total volume 50 L, and the composition was stirred at  $85^{\circ}/70$  mm with continuous addition of the mixture at 23.5 kg/h and II at 13.2 kg/h, addition of polymerization inhibitor, and removal of 1.69 kg/h aqueous phase containing 0.002% I and 0.08% II and 0.18 kg/h organic phase containing I 0.003, isooctane 3.7, and H2O 2.5%, the conversions of I and II being 59.5% and 56.8%, resp., and the selectivities for I and II being 99.2% and 99.12%, resp. L16 ANSWER 43 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 41 ACCESSION NUMBER: 1989:407985 CAPLUS DOCUMENT NUMBER: 111:7985 Preparation of (dialkylamino) alkyl TITLE: (meth) acrylates Hurtel, Patrice; Hazan, Charles; De Champs, Francois; INVENTOR(S): Paul, Jean Michel Norsolor S. A., Fr. PATENT ASSIGNEE(S): Eur. Pat. Appl., 4 pp. SOURCE: CODEN: EPXXDW Patent DOCUMENT TYPE: LANGUAGE: French FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION: DAMENIM NO חאתב KIND DAME ADDITE ATTOM NO

PATENT NO.	KIND	DATE	APPLICATION NO.	DAIL
EP 298867	A2	19890111	EP 1988-401771	19880707
EP 298867	A3	19890308		
EP 298867	B1	19920923		
R: AT, BE, C	H, DE	, ES, GB, GR,	IT, LI, LU, NL, SE	
FR 2617840	A1	19890113	FR 1987-9697	19870708
FR 2617840	B1	19890901		
CA 1303056	A1	19920609	CA 1988-571353	19880707
AT 80868	E	19921015	AT 1988-401771	19880707
ES 2034317	Т3	19930401	ES 1988-401771	19880707
JP 01038047	A2	19890208	JP 1988-170723	19880708
US 4851568	Α	19890725	US 1988-216593	19880708
PRIORITY APPLN. INFO.:			FR 1987-9697	19870708
<del></del>			EP 1988-401771	19880707
OTHER SOURCE(S):	CA	SREACT 111:79	85; MARPAT 111:7985	_

The esters CH2:C(R1)CO2ZNR2R3 (R1 =H, Me; R2, R3 =alkyl, aryl, or form a ring; Z =C1-5 alkylene) are **prepared** by heating Et (meth)acrylate with R3R2NZOH in the presence of Ti(OEt)4 and polymerization **inhibitors**. Thus, heating 660 parts Et acrylate, 267 parts Me2NCH2CH2OH, 1000 ppm phenothiazine, and 0.5 mol% (based on amino **alc.**) Ti(OEt)4 at 90-95° and 0.53 bar with distillation of acrylate-EtOH azeotrope gave 422 parts 2-(dimethylamino)ethyl acrylate.

L16 ANSWER 44 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 42

ACCESSION NUMBER: 1989:515916 CAPLUS

DOCUMENT NUMBER: 111:115916

Preparation of (meth) TITLE:

acrylic acid esters by transesterification

Beranek, Jan; Gutwirth, Karel; Machova, Marta; Benes, INVENTOR(S):

Radek; Kantor, Milan

Czech. PATENT ASSIGNEE(S):

Czech., 5 pp. CODEN: CZXXA9 SOURCE:

Patent DOCUMENT TYPE: Czech LANGUAGE: FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE KIND DATE PATENT NO. \_\_\_\_\_ \_\_\_\_\_ CS 1987-1241 19870225 CS 1987-1241 19870225 CS 259632 B1 19881014 PRIORITY APPLN. INFO.:

CASREACT 111:115916 OTHER SOURCE(S):

The transesterification of CH2:CRCO2R1 (R = H, Me, R1 = Me, Et) with higher aliphatic or cycloaliph. alcs. and glycols is catalyzed by Mg(OMe)2, which is easy to remove as MgCO3. Thus, a mixture of Alfol 1620 (cetyl and stearyl alc.) 434, CH2:CMeCO2Me (I) 194, cyclohexane 333, and phenyl- $\beta$ -naphthylamine (polymerization inhibitor) 1 g was azeotropically dehydrated; a suspension of 0.7 g Mg in 14 g MeOH was added at 70°; and the mixture was refluxed 4 h with separation of MeOH. The residual solution was diluted with 4 g water and treated 15 min with 172 mL/min CO2, which was heated to 95° in a hot washing bottle. Residual CO2 was flushed with air, precipitated MgCO3 was filtered with kiselguhr, and solvents and I were steam-stripped to give 98% cetyl-stearyl methacrylate containing <2% alcoholates and <0.5% I. The filter cake containing 70% MgCO3 was utilized as a fertilizer component.

L16 ANSWER 45 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 43

1988:205252 CAPLUS ACCESSION NUMBER:

108:205252 DOCUMENT NUMBER:

Process for producing unsaturated TITLE:

carboxylates

Nakashima, Sumio; Sogabe, Hideki; Yoshida, Hiroshi; INVENTOR(S):

Okubo, Atsushi

Nippon Shokubai Kagaku Kogyo Co., Ltd., Japan PATENT ASSIGNEE(S):

PCT Int. Appl., 25 pp. SOURCE:

CODEN: PIXXD2

DOCUMENT TYPE: Patent Japanese LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PAT	TENT NO.		KIND	DATE	API	PLICATION NO.	DATE
							1000000
WO	8800180		A1	19880114	WO	1987-JP482	19870707
	W: BR, H	IJ,	KR, US				
	RW: DE, F	'n,	GB, IT				
JP	63017844		A2	19880125	JΡ	1986-159674	19860709
	273060		A1	19880706	ΕP	1987-904331	19870707
EР	273060		B1	19911016			
	R: DE, F	R,	GB, IT				
BR	8707412		Α	19881101	BR	1987-7412	19870707
HU	48863		A2	19890728	HU	1987-3812	19870707
HU	204024		В	19911128			
CA	1290767		A1	19911015	CA	1987-541491	19870707

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CN 1987-105726
                                                        19870709
    CN 87105726
                   Α
                          19880706
    CN 1024190
                     В
                          19940413
                                                        19870709
                                        CS 1987-5239
                     B2
                          19910115
    CS 272223
                                        US 1988-193523 19880309
                          19890523
    US 4833267
                     Α
                                                        19860709
                                     JP 1986-159674
PRIORITY APPLN. INFO.:
                                                        19870707
                                     WO 1987-JP482
```

OTHER SOURCE(S): CASREACT 108:205252

In esterification of (meth)acrylic acid with C1-12 aliphatic alcs., a strongly acidic ion exchange resin catalyst is dispersed in the reaction mixture by a mixer having a stirring power of 0.005-2 kW/m3 reaction solution Thus, 36.7 kg/h mixture solution containing acrylic acid 31.4, 2-ethylhexanol 59.4,

2-ethylhexyl

acrylate (I) 9.1, and H2O 0.1% was fed to a reactor containing 14 L Diaion Pk 208 and 0.05% polymerization **inhibitor**, and the reaction mixture was stirred (0.05 kW/m3) at  $85^{\circ}/70$  mmHg to give a bottom discharge containing 59.99% I with 58.9% conversion (based on acrylic acid) and 99.19% selectivity.

L16 ANSWER 46 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 44

ACCESSION NUMBER:

1988:427631 CAPLUS

DOCUMENT NUMBER:

109:27631

TITLE:

(Meth) acrylate dental material

INVENTOR(S):

Schmitt, Werner; Jochum, Peter; Zahler, Wolf Dietrich; Huebner, Heijo; Holupirek, Manfred; Gasser, Oswald;

Herzig, Christian

PATENT ASSIGNEE(S):

ESPE Fabrik Pharmazeutischer Praeparate G.m.b.H., Fed.

Rep. Ger.

SOURCE:

AB

Ger. Offen., 5 pp.

CODEN: GWXXBX

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 3607331	A1	19870910	DE 1986-3607331	19860306
US 4795823	Α	19890103	US 1987-20902	19870302
JP 62230750	A2	19871009	JP 1987-49874	19870303
JP 07068172	B4	19950726		
EP 235826	A1	19870909	EP 1987-103200	19870306
EP 235826	B1	19900425		
R: AT, BE,	CH, DE	, FR, GB,	LI, LU, NL, SE	
AT 52245	E	19900515	AT 1987-103200	19870306
PRIORITY APPLN. INFO	. :		DE 1986-3607331	19860306
			EP 1987-103200	19870306

The (meth) acrylic acid esters (MO)n(AO2CCH2OCH2CH2OCH2CO2)xA(OM)n [A = alc. radical with C≥4 between the linkage sites; M = CH2:CR1CO; R1 = H, Me; n = 1, 2; x = 0.3-3] are prepared as dental material.

Bis(hydroxymethyl)tricyclo[5.2.1.02,6]decane (T-diol) (196 g) in 400 mL cyclohexane was partially esterified with 89 g triglycolic acid in the presence of 7 g 4-MeC6H4SO3H with the removal of 18 g H2O. The free OH groups were esterified with 129 g CH2:CMeCO2H in the presence of polymerization inhibitors to give an ester mixture consisting of 32% T-diol dimethacrylate, 38% triglycolic acid bis[T-methacrylate] and 30% HO(TOCOCH2CCH2CCH2CO2)2-3TOH bis(methacrylate). A solution was prepared of 50 parts by weight bis(acryloyloxymethyl)tricyclo[5.2.1.02, 6]decane and 50 parts of the ester mixture and 0.15% camphorquinone and 1.5%

N, N-dimethylaminoethyl methacrylate added. Sep., a powder was prepared containing 112 parts silanized SiO2 granulate, 72 parts silanized pyrogenic SiO2 and 3 g CaF2. A dental composition was prepd . by mixing 113 g of the powder with 80 mL of the solution Polymerization was carried out by light exposure as usual.

L16 ANSWER 47 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 45

ACCESSION NUMBER:

1987:157005 CAPLUS

DOCUMENT NUMBER:

106:157005

TITLE:

(Meth)acrylate esters

INVENTOR(S):

Fujii, Masahiko; Hashino, Shizuo

PATENT ASSIGNEE(S):

Nippon Oil Co., Ltd., Japan

SOURCE:

Jpn. Kokai Tokkyo Koho, 6 pp. CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

DATE APPLICATION NO. DATE KIND DATE PATENT NO. \_\_\_\_\_ JP 1985-82267 19850419
.TP 1985-82267 19850419 JP 61243046 A2 19861029 JP 1985-82267 PRIORITY APPLN. INFO.:

OTHER SOURCE(S):

CASREACT 106:157005

(Meth)acrylate esters, useful as monomers for resins, coating materials, and adhesives, are prepared by esterification of alcs. with excess (meth) acrylic acid in the presence of acid catalysts and polymerization inhibitors. The

reaction mixture is neutralized with aqueous alkali solns., inorg. acids are added to pH  $\leq$ 4, and the unesterified acid is extracted with aliphatic or petroleum ethers from the aqueous layer. CH2:CMeCO2H was treated with Me(CH2)170H in the presence of 4-MeC6H4SO3H and hydroquinone at 110-115°/200-300 torr for 4.5 h, then the solution was neutralized with 17.5% aqueous NaOH, and separated from the oil layer to give 95.8% CH2:CMeCO2(CH2)17Me. The aqueous layer was extracted with Et20 to recover the unesterified starting acid in 99.5% purity and reduce the COD of the effluent by 79.5%.

L16 ANSWER 48 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 46

ACCESSION NUMBER:

1987:67824 CAPLUS

DOCUMENT NUMBER:

106:67824

TITLE:

(Meth)acrylate esters

INVENTOR(S): PATENT ASSIGNEE(S): Hashino, Shizuo; Tanabe, Tatsuhei Nippon Oils & Fats Co., Ltd., Japan

SOURCE:

Jpn. Kokai Tokkyo Koho, 4 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

inhibitor, followed by esterifying residual (

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE		APPLICATION NO.	DATE	
JP 61176555	A2	19860808		JP 1985-14447	19850130	
JP 06080033	B4	19941012				
PRIORITY APPLN. INFO	. :		JР	1985-14447	19850130	
AB (Meth)acrylate esters are prepared by esterifying higher						
alcs. with excess (meth)acrylic acid						

in presence of both an esterifying catalyst and a polymerization

meth) acrylic acid with lower alcs., and fractionating the mixed (meth) acrylate esters. This method gives higher alc. esters in high yield and purity with efficient recovery of the excess acid. Thus, heating H2C:CMeCO2H (I) 103.3, octadecyl alc. 270.0, 4-MeC6H4SO3H 3.5, and hydroquinone 0.11 g at 110-115°/200-600 torr for 4.5 h gave a 99.2% conversion of the alc. and 13.6 g residual I. Heating the mixture with 30.3 g MeOH at 80° for 3 h and distilling the final mixture at 46°/160 torr gave a distillate containing 15.1 g Me methacrylate corresponding to a 95.2% recovery of residual I, and a residue containing 324.6 g octadecyl methacrylate corresponding to a 96.0% recovery of the alc.

L16 ANSWER 49 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 47

ACCESSION NUMBER: 1987:120368 CAPLUS

DOCUMENT NUMBER: 106:120368

TITLE: Alkyl methacrylate preparation

INVENTOR(S): Dietrich, Gerhard; Nestler, Gerhard; Ruckh, Peter;

Herzog, Reinhard

PATENT ASSIGNEE(S): BASF A.-G., Fed. Rep. Ger.

SOURCE: Ger. Offen., 8 pp.

CODEN: GWXXBX

DOCUMENT TYPE: Patent LANGUAGE: German

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 3518482	A1	19861127	DE 1985-3518482	19850523
US 4675436	Α	19870623	US 1986-857052	19860429
JP 61271247	A2	19861201	JP 1986-108733	19860514
JP 06017337	B4	19940309		
EP 202610	A2	19861126	EP 1986-106633	19860515
EP 202610	A3	19870128		
EP 202610	B1	19880928		
R: BE, DE,	FR, GE	B, IT, NL		

PRIORITY APPLN. INFO.: DE 1985-3518482 19850523

OTHER SOURCE(S): CASREACT 106:120368

AB Alkyl (meth) acrylates are prepared from (meth) acrylic acid and C6-20 alcs. in the presence

of 0.1-5% strong acids, polymerization inhibitors, and an O-containing gas (1-20 vol%) at 80-150°, for the elimination of H2O. The (

meth) acrylic acid-alc. ratio is

1:0.8-1.2. To a 1st reactor was fed methacrylic acid 190, H2SO4 5, phenothiazine 0.3, and n-octanol 280 parts/h. To a 2nd reactor was fed the mixture from the 1st reactor with 36 parts/h n-octanol, and 150,000 volume parts air. At 120°, 40 parts/h H2O was eliminated. The product (470 parts/h) contained n-octyl methacrylate 90.5, methacrylic acid 0.7, octanol 5.8, and high-boiling side products (traces of dioctyl ether and olefins) 0.5%.

L16 ANSWER 50 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 48

ACCESSION NUMBER: 1984:121755 CAPLUS

DOCUMENT NUMBER: 100:121755

TITLE: Acrylate or methacrylate esters
PATENT ASSIGNEE(S): Yokkaichi Chemical Co., Ltd., Japan

Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

SOURCE:

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 58174346 A2 19831013 JP 1982-57285 19820408

JP 01042255 B4 19890911

PRIORITY APPLN. INFO.: JP 1982-57285

AB Acrylic acid (I) [79-10-7] or methacrylic acid [79-41-4] is

esterified with alcs. in the presence of

esterification catalysts, antioxidants, and Cu or Cu salts (polymerization inhibitors), and the products are treated with pyrophosphoric acid (II), II salts, 1-hydroxyethylidenebis(phosphonic acid) (III) [2809-21-4], or III salts to remove Cu ions. Thus, a mixture of 1.10 equiv I, 1.0 equiv 1,1'-isopropylidenebis[4-[2-(2hydroxethoxy)ethoxy]benzene] [27697-57-0], 3.4% (based on I) 4-MeC6H4SO3H, 100 ppm (on I) powdered Cu, 100° (on I) PhMe, and 0.4% NaH2PO2 was heated to reflux, removing H2O until the acid number reached a constant value and the product cooled, neutralized with dilute NaOH, treated with 0.55% (on I) III, separated from the aqueous phase, and stripped of solvents

in vacuo at 50-60° to give an ester [56361-55-8] having APHA color number 30 and containing only a trace of Cu ion.

L16 ANSWER 51 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 49

ACCESSION NUMBER: 1981:192923 CAPLUS

94:192923 DOCUMENT NUMBER:

Monomers for polymeric depressing additives for TITLE:

high-paraffin oils

Minkov, V. A.; Sopina, V. E.; Zakordonets, O. P.; INVENTOR(S):

Sergeev, V. P.; Gaevoi, G. M.; Khmelnitskii, A. G.; Lubenets, E. G.; Saveleva, N. I.; Skovorodnikov, Yu.

Α.

PATENT ASSIGNEE(S): USSR

U.S.S.R. From: Otkrytiya, Izobret., Prom. Obraztsy, SOURCE:

Tovarnye Znaki 1981, (9), 88.

CODEN: URXXAF

DOCUMENT TYPE: Patent Russian LANGUAGE:

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE PATENT NO. KIND DATE \_\_\_\_\_ \_\_\_\_\_\_ SU 1978-2694189 19781206 SU 810668 A1 19810307 SU 1978-2694189 19781206 PRIORITY APPLN. INFO.:

Monomers for pour-point-depressant polymers were prepared by simultaneous esterification of acrylic and methacrylic acid, in a 2-5:1 mol. ratio, with a C18-26 alc. fraction in toluene. The reaction was carried out at 100-20° using toluenesulfonic acid [104-15-4] as catalyst and hydroquinone [123-31-9] as polymerization inhibitor.

L16 ANSWER 52 OF 52 CAPLUS COPYRIGHT 2004 ACS on STN DUPLICATE 50

ACCESSION NUMBER: 1981:47993 CAPLUS

DOCUMENT NUMBER: 94:47993

(Meth) acrylic acid TITLE:

esters

Schuster, Karl Ernst; Rosenkranz, Hans Juergen; Griehsel, Bernd INVENTOR(S):

number of

CA SUBSCRIBER PRICE

Bayer A.-G., Fed. Rep. Ger. PATENT ASSIGNEE(S):

Ger. Offen., 19 pp. SOURCE:

CODEN: GWXXBX

DOCUMENT TYPE: Patent LANGUAGE: German

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

ALNU DATE APPLICATION NO. DATE PATENT NO. KIND DATE \_\_\_\_\_ DE 1979-2913218 19790403 19801023 DE 2913218 A1 PRIORITY APPLN. INFO.: DE 1979-2913218 19790403

(Meth) acrylates of saturated aliphatic alcs. containing 2-4 OH groups, their ethoxylation products, or unsubstituted or C1-3 alkyl mono- or disubstituted 5-(hydroxymethyl)-1,3-dioxanes are prepared by esterification in the presence of 0.001-5% organic phosphite catalyst and 0.01-0.3% mono- or diphenol as polymerization inhibitor. Thus, acrylic acid [79-10-7] 3.5, ethoxylated trimethylolpropane [50586-59-9] 5.36, cyclohexane 2.7, and H2SO4 0.073 kg were mixed with 0.006 kg (EtO)3P [122-52-1] and 2,5-di-tert-butylhydroquinone (I) [88-58-4] and esterified 17 h at 82°, while 10 L/h (EtO)3P-saturated air and 10 L/h I-saturated air were passed through the mixture, so that the acid

the reaction mixture was 12. The mixture was cooled, mixed with (EtO)3P 0.005, I 0.0024, and toluhydroquinone [95-71-6] 0.002 kg, distilled free of cyclohexane, and distilled for 4 h at 50 mbar/105° while 50 L/h (EtO) 3P-saturated air was passed through the mixture The final product [28961-43-5] had acid number 2.5, iodine color number 0-1, and viscosity 120 mPa-s. A control prepared without (EtO)3P had iodine color number 2-3 and viscosity 190 mPa-s.

=> log y SINCE FILE TOTAL COST IN U.S. DOLLARS SESSION ENTRY 195.12 195.33 FULL ESTIMATED COST SINCE FILE TOTAL DISCOUNT AMOUNTS (FOR OUALIFYING ACCOUNTS) SESSION ENTRY -38.22 -38.22

STN INTERNATIONAL LOGOFF AT 13:39:59 ON 28 JUL 2004